



# Acid, Bases And Salts

- These are the substances which have sour taste.
- They turn blue litmus solution red.
- They give  $H^+$  ions in aqueous solution.
- The term 'acid' has been derived from the Latin word, acidus, which means sour.

**Strong Acids :**  $\text{HCl}$ ,  $\text{H}_2\text{SO}_4$ ,  $\text{HNO}_3$

**Weak Acids :**  $\text{CH}_3\text{COOH}$ , Oxalic acid, Lactic acid

**Concentrated Acid :** Having more amount of acid + less amount of water

**Dilute Acid :** Having more amount of water + less amount of acid

- These are the substances which are bitter in taste and soapy in touch.
- They turn red litmus solution blue.
- They give  $\text{OH}^-$  ions in aqueous solution.

**Strong Bases :**  $\text{NaOH}$ ,  $\text{KOH}$ ,  $\text{Ca(OH)}_2$ .

**Weak Bases :**  $\text{NH}_4\text{OH}$

**Alkalis :** These are bases which are soluble in water [NaOH, KOH, Ca(OH)<sub>2</sub>].

**These are the compounds formed from reaction of acid and base.**

# Topic 10 Acids Bases And Salts Practice Questions

**Ralph Knupp Watkins**



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