



## Chapter - 2

# Acid, Bases And Salts

### ACIDS :

- These are the substances which have sour taste.
- They turn blue litmus solution red.
- They give  $H^+$  ions in aqueous solution.
- The term 'acid' has been derived from the Latin word, acidus, which means sour.

**Strong Acids** :  $HCl$ ,  $H_2SO_4$ ,  $HNO_3$

**Weak Acids** :  $CH_3COOH$ , Oxalic acid, Lactic acid

**Concentrated Acid** : Having more amount of acid + less amount of water

**Dilute Acid** : Having more amount of water + less amount of acid

### BASES :

- These are the substances which are bitter in taste and soapy in touch.
- They turn red litmus solution blue.
- They give  $OH^-$  ions in aqueous solution.

**Strong Bases** :  $NaOH$ ,  $KOH$ ,  $Ca(OH)_2$

**Weak Bases** :  $NH_4OH$

**Alkalies** : These are bases which are soluble in water [ $NaOH$ ,  $KOH$ ,  $Ca(OH)_2$ ].

### SALTS :

These are the compounds formed from reaction of acid and base.

# Topic 10 Acids Bases And Salts Practice Questions

**Ralph Knupp Watkins**

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