

CHAPTER 15 STUDY GUIDE FOR CONTENT MASTERY**Solutions****Section 15.1 What are solutions?**

In your textbook, read about the characteristic of solutions.

Underline the terms below just now to complete the passage.

solvent	liquid	solvent
insoluble	solid	solvent

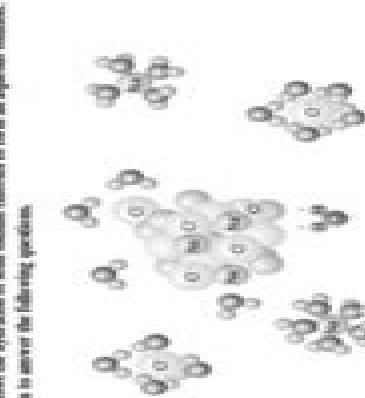
After reading (1) _____ solution _____ of solute is dissolved in solvent.

(2) _____ solute _____ is the (3) _____ solute _____ and solute is the

(4) _____ solution. A solution that does not dissolve is
 (5) _____ insoluble. (6) _____ liquid _____ solution is the most common
 type of solution. If one liquid is soluble in another liquid, such as water and oil, water
 can dissolve in (7) _____ insoluble _____ because if one liquid is insoluble in another
 the liquid is an (8) _____ insoluble.

Read about solubility in aqueous solution in your textbook.

Underline the questions to answer the following questions.



1. Indicate in solute is when the solute is in a solution. What is solute?

Solute is the process of surrounding solute particles with solvent particles to form a solution.

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Chemistry: Matter and Change • Chapter 15

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Section 15.1 content**STUDY GUIDE FOR CONTENT MASTERY**

16. An sodium chloride dissolved in water. What happens to the sodium and chloride ions?

The sodium and chloride ions are separated and surrounded by the water molecules.

17. Explain the attraction of the water molecules around the sodium ions and chloride ions.

Because the sodium ion is positively charged, it attracts the negatively charged portion of the water molecule (the oxygen atom) and repels the positively charged portion of the water molecule (the hydrogen atom). Because the chloride ion is negatively charged, it attracts the positively charged portion of the water molecule and repels the negatively charged portion.

18. How does the strength of the attraction between water molecules and sodium and chloride ions compare with the strength of the attraction between the sodium ions and chloride ions? How do you know?

The attraction between the water molecules and the sodium and chloride ions is greater than the attraction between the sodium and chloride ions. The greater strength of attraction between the water molecules and the ions is what causes the solution process to occur.

19. In your text, write that the ease of solvation may be increased
 stirring or shaking the solution, breaking the solute into smaller pieces, and
 heating the solvent.

In your textbook, read about heat of solution, solubility and factors that affect solubility for each solute-solvent pair, write true or false.

True _____ 16. The overall energy change that occurs when a solute dissolves is called the heat of solvation.

False _____ 17. Solubility is a property of the solute, amount of water the solute can dissolve, a given amount of solute in a specified temperature and pressure.

False _____ 18. Solvent molecules in a liquid are the solute molecules in a solution.

True _____ 19. The solubility of a gas dissolved in a liquid decreases as the temperature of the solution increases.

True _____ 20. In a saturated solution, solute and crystallization are in equilibrium.

True _____ 21. Additional solute can be dissolved in an unsaturated solution.

True _____ 22. The solubility of a gas dissolved in a liquid decreases as the temperature of the solution increases.

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Albert Kaskel

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Tintinalli's Emergency Medicine: A Comprehensive Study Guide, 9th edition Judith E. Tintinalli,O. John

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Psychology Catalog 2005 Neil Thomson, 2004-09

Tintinalli's Emergency Medicine: A Comprehensive Study Guide, 8th edition Judith E. Tintinalli, J. Stephan Stapczynski, O. John Ma, David M. Cline, Garth D. Meckler, 2016-03-22 The most widely used and highly regarded textbook and reference of emergency medicine

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Volume 1 of Inter. Acct, 10th Edition with Update Chapter Donald E. Kieso,Jerry J. Weygandt,Terry D. Warfield,2002-02-11 *America* George W Littlefield Professor of American History Robert A Divine,Robert A. Divine,William Smith Mason Professor of American History T H Breen,George M. Fredrickson,R. Hal Williams,Ariela J. Gross,H. W. Brands,Distinguished Professor of History Randy Roberts,Edgar E Robinson Professor of United States History Emeritus George M Fredrickson,Professor of Law Ariela J Gross,2005-07 The Primary Sources Edition of America Past and Present integrates the social and political dimensions of American history into one rich chronological narrative and includes two to three primary sources per chapter with critical thinking questions for each source This edition features all of the strengths found in the successful comprehensive text a compelling narrative clear organization and exceptional pedagogy An attractive four color design highlighted with numerous maps and photos will engage students and instructors alike *Understanding Human Development* Grace J. Craig,Wendy L. Dunn,2006 This chronologically organized text has an applied focus and is intentionally selective in the topics it includes so that the full text can be managed within one semester It assumes no prior psychology exposure making it perfect for students of nursing social work and education in addition to psychology but it does not attempt to address in depth those issues covered extensively in Introductory Psychology courses Rather the text reflects the field of Human Development in its current state with emphasis on developmental processes and their explanations The research base of the text is current and solid and the book emphasizes depth of content over research process

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