

10.1 Periodic Trends



10.1 PERIODIC TRENDS: FROM CONCEPT TO APPLICATION

Section 10.1: Introduction

In your notebook, read aloud the modern periodic table.

Copy the letter-headers in the first row on the left below from the periodic table to complete the table on the right.

Atomic Mass	_____
Atomic Number	_____
Electron Configuration	_____
Chemical Name	_____
Chemical Symbol	_____

Put each item in Column A, under the letter of the matching item in Column B.

Column A

- _____ 16. It increases in the periodic table.
- _____ 17. It decreases in the periodic table.
- _____ 18. Group 1 elements.
- _____ 19. Elements that are shiny and conduct electricity.
- _____ 20. Group 17 elements.

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write each of the statements in terms of the statement in italics, changing the underlined word as shown in smaller letters.

- _____ 21. There are two representative elements of phosphorus.
- _____ 22. About three times heavier of the elements in the periodic table are transition.
- _____ 23. Group 1A elements except for hydrogen are located on the left of metals.
- _____ 24. Group 2A elements are the alkaline earth metals.
- _____ 25. Group 7A elements are highly reactive nonmetals known as halogens.
- _____ 26. Group 8A elements are inert nonreactive elements known as noble gases.
- _____ 27. Alkaloids have properties of both metals and more reactive metals.

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Carol Ann Tomlinson



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Table of Contents Study Guide For Content Mastery Answers Workbook

1. Understanding the eBook Study Guide For Content Mastery Answers Workbook
 - The Rise of Digital Reading Study Guide For Content Mastery Answers Workbook
 - Advantages of eBooks Over Traditional Books
2. Identifying Study Guide For Content Mastery Answers Workbook
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Study Guide For Content Mastery Answers Workbook
 - User-Friendly Interface
4. Exploring eBook Recommendations from Study Guide For Content Mastery Answers Workbook
 - Personalized Recommendations
 - Study Guide For Content Mastery Answers Workbook User Reviews and Ratings

- Study Guide For Content Mastery Answers Workbook and Bestseller Lists
- 5. Accessing Study Guide For Content Mastery Answers Workbook Free and Paid eBooks
 - Study Guide For Content Mastery Answers Workbook Public Domain eBooks
 - Study Guide For Content Mastery Answers Workbook eBook Subscription Services
 - Study Guide For Content Mastery Answers Workbook Budget-Friendly Options
- 6. Navigating Study Guide For Content Mastery Answers Workbook eBook Formats
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 - Study Guide For Content Mastery Answers Workbook Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Study Guide For Content Mastery Answers Workbook
 - Highlighting and Note-Taking Study Guide For Content Mastery Answers Workbook
 - Interactive Elements Study Guide For Content Mastery Answers Workbook
- 8. Staying Engaged with Study Guide For Content Mastery Answers Workbook
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Study Guide For Content Mastery Answers Workbook
- 9. Balancing eBooks and Physical Books Study Guide For Content Mastery Answers Workbook
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Study Guide For Content Mastery Answers Workbook
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Study Guide For Content Mastery Answers Workbook
 - Setting Reading Goals Study Guide For Content Mastery Answers Workbook
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Study Guide For Content Mastery Answers Workbook
 - Fact-Checking eBook Content of Study Guide For Content Mastery Answers Workbook
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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