

**Section 8.1** *continued*

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

|       |       |
|-------|-------|
| _____ | _____ |
| _____ | _____ |
| _____ | _____ |
| _____ | _____ |
| _____ | _____ |

|                        |       |
|------------------------|-------|
| Atomic Mass            | _____ |
| Atomic Number          | _____ |
| Electron Configuration | _____ |
| Chemical Name          | _____ |
| Chemical Symbol        | _____ |

Put each item in Column A, either the letter of the matching item in Column B.

**Column A**

- \_\_\_\_\_ 18. It appears in the periodic table.
- \_\_\_\_\_ 19. It has no line periodic table.
- \_\_\_\_\_ 20. Group 1 elements.
- \_\_\_\_\_ 21. Elements that are strong oxidizing agents.
- \_\_\_\_\_ 22. Group 8 elements.

**Column B**

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write *yes* if the statement is true, if the statement is false, change the underlined word or phrase to make it true.

- \_\_\_\_\_ 23. There are two representative elements of elements.
- \_\_\_\_\_ 24. Most elements have lines in the periodic table for metals.
- \_\_\_\_\_ 25. Group 1 elements except for hydrogen are metals in the solid state.
- \_\_\_\_\_ 26. Group 2 elements are the alkaline earth metals.
- \_\_\_\_\_ 27. Group 1 elements are highly reactive elements known as alkalis.
- \_\_\_\_\_ 28. Group 8 elements are very reactive elements known as noble gases.
- \_\_\_\_\_ 29. Elements have properties of both metals and nonmetals.

# Study Guide For Content Mastery Periodic Trends

**T. A. Lamke, Herbert M. Silvey**



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





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