

10.1 Periodic Trends



10.1 PERIODIC TRENDS: FROM CONCEPT TO APPLICATION

Section 10.1 – continued

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

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| | |
|------------------------|-------|
| Atomic Mass | _____ |
| Atomic Number | _____ |
| Electron Configuration | _____ |
| Chemical Name | _____ |
| Chemical Symbol | _____ |

Put each item in Column A, either the letter of the matching item in Column B.

Column A

- _____ 1. It increases in the periodic table.
- _____ 2. It decreases in the periodic table.
- _____ 3. Group 1 elements.
- _____ 4. Elements that are shiny and conduct electricity.
- _____ 5. Group 17 elements.

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write each of the statements in terms of the statement in bold; change the underlined word or phrase to make it true.

- _____ 1. There are two representative elements of elements.
- _____ 2. There are two periods of the elements in the periodic table.
- _____ 3. Group 1 elements are highly reactive elements in the periodic table.
- _____ 4. Group 17 elements are the most reactive elements.
- _____ 5. Group 1 elements are highly reactive elements in the periodic table.
- _____ 6. Group 17 elements are highly reactive elements in the periodic table.
- _____ 7. Group 1 elements are highly reactive elements in the periodic table.
- _____ 8. Group 17 elements are highly reactive elements in the periodic table.
- _____ 9. Group 1 elements are highly reactive elements in the periodic table.
- _____ 10. Group 17 elements are highly reactive elements in the periodic table.

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