

Name _____ Date _____ Class _____

CHAPTER 4 STUDY GUIDE FOR CONTENT MASTERY**The Structure of the Atom****Section 4.1 Early Theories of Matter***In your textbook, read about the philosophers, John Dalton, and defining the atom.*For each statement below, write *true* or *false*.

- false** _____ 1. Ancient philosophers carefully performed controlled experiments.
- true** _____ 2. Philosophers formulated explanations about the nature of matter based on their own experiences.
- true** _____ 3. Both Democritus and Dalton suggested that matter is made up of atoms.
- true** _____ 4. Dalton's atomic theory stated that atoms separate, combine, or rearrange in chemical reactions.
- false** _____ 5. Dalton's atomic theory stated that matter is mostly empty space.
- false** _____ 6. Dalton was correct in thinking that atoms could not be divided into smaller particles.
- true** _____ 7. Dalton's atomic theory stated that atoms of different elements combine in simple whole-number ratios to form compounds.
- true** _____ 8. Dalton thought that all atoms of a specific element have the same mass.
- false** _____ 9. Democritus proposed that atoms are held together by chemical bonds, but no one believed him.
- true** _____ 10. Dalton's atomic theory was based on careful measurements and intensive research.

false _____ 11. There are no instruments powerful enough to measure atoms so that they can be seen.

true _____ 12. The smallest particle of an element that retains the properties of that element is called an atom.

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CHAPTER 4 STUDY GUIDE FOR CONTENT MASTERY**Section 4.2 Subatomic Particles and the Nuclear Atom***In your textbook, read about discovering the electron and the nuclear atom.*

For each item in Column A, write the letter of the matching item in Column B.

Column A	Column B
_____ 1. Proposed the nuclear atomic model	a. Thomson
_____ 2. Determined the mass-to-charge ratio of an electron	b. Millikan
_____ 3. Calculated the mass of an electron	c. Rutherford

Draw and label a diagram of each atomic model.

a. plum pudding model

Drawing should look like a ball of chocolate chip cookie dough. The chocolate chips should be labeled with negative charge or as electrons. The dough should be labeled as evenly distributed positive charges.

b. nuclear atomic model

Drawing should look like a peach with a pit. The pit should be labeled nucleus and should include labeled protons and neutrons. The outer circle of the peach should be labeled electrons.

In your textbook, read about the discovery of protons and neutrons.

Complete the following table of protons, electrons, and neutrons characteristics.

Particle	Symbol	Location	Relative Charge	Relative Mass
1. Proton	p^+	In the nucleus	$1+$	1
2. Neutron	n^0	In the nucleus	0	1
3. Electron	e^-	In the space surrounding the nucleus	$1-$	1/1840

Study Guide Answers Nuclear Change

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Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03
The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section, and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts, and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem. Every concept in the text is clearly illustrated with one or more step-by-step examples. Making it Real essays have been updated to present timely and engaging real-world applications, emphasizing the relevance of the material they are learning. This edition continues the end-of-chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter.

Study Guide to Accompany Calculus for the Management, Life, and Social Sciences Clyde Metz, 1984-01-01
Study Guide to Accompany Calculus for the Management Life and Social Sciences

Student Study Guide and Solutions Manual to Accompany General, Organic, and Biochemistry Katherine J. Denniston, Katherine J. Denniston, 2006

Student Study Guide and Solutions Manual to accompany General Organic and Biological Chemistry, 1e Kenneth W. Raymond, 2005-10-07
Finally, readers have a shorter, less intimidating introduction to general organic and biological chemistry. Not only is Raymond's text concise, it also takes an integrated approach to presenting important topics in a way that makes the material easier to understand. In this approach, similarities can be exploited and concepts reinforced. The result is that readers see the strong connections that exist between these three branches of chemistry.

Ebook: Chemistry: The Molecular Nature of Matter and Change Silberberg, 2015-01-16
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Study Guide to Accompany Garrett & Hough's Brain & Behavior: An Introduction to Behavioral Neuroscience Bob Garrett, Gerald Hough, 2017-10-13
Revised to accompany Bob Garrett's best seller *Brain Behaviour: An Introduction to Biological Psychology*, this fully updated Student Study Guide provides additional opportunities for student practice and self-evaluation.

The Study Guide for Developing Person Through Childhood and Adolescence Richard O. Straub, 2008-12-18
Each chapter includes a review of key concepts, guided study questions, and section reviews that encourage students' active participation in the learning process. Two practice tests and a challenge test help them assess their mastery of the material. Applications and observational activities are also included.

The Study Guide for Developing Person Through the Life Span Kathleen Stassen Berger, 2007-12-24
This seventh edition comes with a significant revision of cognitive development through childhood, revised and updated chapters on adolescence, and more attention to emerging and early adulthood.

Holt Chemistry Holt Rinehart & Winston, 2003-01-24

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Guide for Introduction to Maternity and Pediatric Nursing - E-Book Gloria Leifer, 2015-01-29 Leifer's Study Guide for Introduction to Maternity and Pediatric Nursing is a rich resource that enhances your comprehension and application of material from every chapter of the textbook. With a variety of learning activities, critical thinking exercises, case studies, and review questions, this study guide reinforces your mastery of the essential maternity and pediatric nursing concepts and skills you need to pass the NCLEX PN Examination and succeed in your nursing career. NCLEX exam style multiple choice review questions at the end of each chapter test basic chapter knowledge as well as ask for appropriate nursing actions, what the nurse should expect in terms of medical care of the patient, and what complications the patient is at risk of developing. Case Studies and Applying Knowledge activities provide additional opportunities to apply learned information to clinical care. Thinking Critically activities require you to apply what you've learned in the textbook to new situations and draw conclusions based on that knowledge. Learning activities include matching, labeling, and completion exercises to help you learn basic concepts of maternity and pediatric nursing. Crossword puzzles in selected chapters are a nice way to change up the format when testing comprehension of terms and concepts. Answer key is located on the student's Evolve website. Several Thinking Critically sections and case studies with critical thinking questions have been added. Alternate item format questions are included in the review questions for the NCLEX Examination.

Study Guide Joseph J. Boyle, Douglas C. Giancoli, 1998-2000
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Pamphlets on Protozoology (Kofoid Collection), 1912
Study Guide Kelly Bouas-Henry, 2005 By Kelly Bouas Henry Missouri Western State College and Douglas A. Bernstein the Study Guide employs numerous techniques to help students learn. Each chapter, including the optional Industrial/Organizational Psychology chapter, contains a detailed outline, a key terms section that presents fresh examples and learning aids, plus a fill-in-the-blank test, learning objectives, a concepts and exercises section that shows students how to apply their knowledge of psychology to everyday issues and concerns, a critical thinking exercise, and personal learning activities. In addition, each chapter concludes with a two-part self-quiz consisting of 40 multiple choice questions. An answer key tells the student not only which response is correct but also why each of the other choices is wrong, and quiz analysis tables enable students to track patterns to their wrong answers, either by topic or by type of question: definition, comprehension, or application.

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Study Guide Plus Sociology Barbara Marliene Scott, 2000-09
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Student Topic Outline and Study Guide to Accompany World of Chemistry, Essentials, Joesten/Netterville/Wood James O. Schreck, 1993

Telecourse Study Guide for Sociology Pearson, James M. Henslin, Shelly Dutchin, 2005-04 This student manual is for students who are using Sociology: A Down to Earth Approach with the twenty-two video programs in the Exploring Society series from Dallas TeleLearning. Each section coordinates reading and video assignments and includes summaries, learning

objectives video outlines key terms and people and student application projects The Guide also provides a self test section containing multiple choice true false fill in matching and essay questions that correlate to the video programs as well as Henslin s text **Prentice Hall Chemistry** ,2000

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