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**CHAPTER 4** STUDY GUIDE FOR CONTENT MASTERY

# The Structure of the Atom

**Section 4-1 Early Theories of Matter**

In your textbook, read about the phlogiston, John Dalton, and atomic theories.

For each statement below, write **True** or **False**.

**True** \_\_\_\_\_ 1. Ancient philosophers quickly performed controlled experiments.

**True** \_\_\_\_\_ 2. John Dalton formulated a hypothesis about the nature of matter based on his own principles.

**True** \_\_\_\_\_ 3. John Dalton and Thomas Young proposed that matter is made up of atoms.

**True** \_\_\_\_\_ 4. Dalton's atomic theory stated that atoms are small, uniform, and indestructible.

**False** \_\_\_\_\_ 5. Dalton's atomic theory stated that atoms are empty spaces.

**False** \_\_\_\_\_ 6. Dalton's atomic theory stated that atoms could not be divided into smaller particles.

**False** \_\_\_\_\_ 7. Dalton's atomic theory stated that atoms of different elements combine in simple whole number ratios to form compounds.

**True** \_\_\_\_\_ 8. Dalton thought that all atoms of a specific element have the same mass.

**False** \_\_\_\_\_ 9. Dalton proposed his atoms are held together by chemical bonds, but in our everyday life.

**False** \_\_\_\_\_ 10. Dalton's atomic theory was based on useful measurements and accurate research.

**False** \_\_\_\_\_ 11. There are no electrons present enough to supply atoms to their full capacity.

**True** \_\_\_\_\_ 12. The number of protons in an atom is equal to the number of neutrons in an atom.

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**CHAPTER 4** STUDY GUIDE FOR CONTENT MASTERY**Section 4-2 Subatomic Particles and the Nuclear Atom**

In your textbook, read about identifying the electrons and the nuclear atom.

For each item in Column A, write the letter of the matching item in Column B.

**Column A**

**a** \_\_\_\_\_ 1. Proposed the nuclear atomic model

**b** \_\_\_\_\_ 2. Determined the mass-to-charge ratio of an atom

**c** \_\_\_\_\_ 3. Observed the mass of an atom

**d** \_\_\_\_\_ 4. Determined the mass of an electron

Draw and label a diagram of each atomic model.

**a** \_\_\_\_\_ 4. **John Dalton's model**  
Drawing should look like a ball of chocolate chip cookie dough. The chocolate chip should be labeled with negative charge or as electrons. The dough should be labeled as evenly distributed positive charges.

**b** \_\_\_\_\_ 3. **Thomson's model**  
Drawing should look like a ball of chocolate chip cookie dough.

**c** \_\_\_\_\_ 1. **Miller's model**  
Drawing should look like a patch with a pit. The pit should be labeled nucleus and should include labeled protons and neutrons. The outer edge of the patch should be labeled electrons.

**d** \_\_\_\_\_ 2. **Rutherford's model**  
Drawing should look like a patch with a pit. The pit should be labeled nucleus and should include labeled protons and neutrons. The outer edge of the patch should be labeled electrons.

In your textbook, read about the discovery of protons and neutrons.

Complete the following table of protons, neutrons, and neutron characteristics.

Particle	Symbol	Location	Relative Charge	Relative Mass
1. Proton	$p^+$	In the nucleus	1+	1
2. Neutron	$n^0$	In the nucleus	0	1
3. Electron	$e^-$	In the space surrounding the nucleus	1-	1/1836

Study Guide for Content Mastery

**CHAPTER 4** Chemistry: Matter and Change

Study Guide for Content Mastery

# Study Guide Answers Nuclear Change

**Holt Rinehart & Winston**

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**Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual** Leo J. Malone, Theodore O. Dolter, 2012-01-03

The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem. Every concept in the text is clearly illustrated with one or more step by step examples. Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning. This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter.

**Study Guide to Accompany Calculus for the Management, Life, and Social Sciences** Clyde Metz, 1984-01-01 Study Guide to Accompany Calculus for the Management Life and Social Sciences

[Student Study Guide and Solutions Manual to Accompany General, Organic, and Biochemistry](#)

Katherine J. Denniston, Denniston Katherine, 2006 **Student Study Guide and Solutions Manual to accompany**

**General Organic and Biological Chemistry, 1e** Kenneth W. Raymond, 2005-10-07 Finally readers have a shorter less intimidating introduction to general organic and biological chemistry. Not only is Raymond's text concise, it also takes an integrated approach to presenting important topics in a way that makes the material easier to understand. In this approach, similarities can be exploited and concepts reinforced. The result is that readers see the strong connections that exist between these three branches of chemistry.

*Ebook: Chemistry: The Molecular Nature of Matter and Change* Silberberg, 2015-01-16

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[Behavior: An Introduction to Behavioral Neuroscience](#) Bob Garrett, Gerald Hough, 2017-10-13 Revised to accompany Bob

Garrett's best seller Brain Behaviour An Introduction to Biological Psychology this fully updated Student Study Guide

provides additional opportunities for student practice and self evaluation.

*The Study Guide for Developing Person Through Childhood and Adolescence* Richard O. Straub, 2008-12-18 Each chapter includes a review of key concepts, guided study questions and section reviews that encourage students active participation in the learning process. Two practice tests and a challenge test help them assess their mastery of the material. Applications and observational activities are also included.

[The Study Guide for Developing Person Through the Life Span](#) Kathleen Stassen Berger, 2007-12-24 This seventh edition comes with a significant revision of cognitive development through childhood revised and updated chapters on adolescence

and more attention to emerging and early adulthood **Holt Chemistry** Holt Rinehart & Winston, 2003-01-24 **Study**

**Guide for Introduction to Maternity and Pediatric Nursing - E-Book** Gloria Leifer,2015-01-29 Leifer's Study Guide for Introduction to Maternity and Pediatric Nursing is a rich resource that enhances your comprehension and application of material from every chapter of the textbook. With a variety of learning activities critical thinking exercises case studies and review questions this study guide reinforces your mastery of the essential maternity and pediatric nursing concepts and skills you need to pass the NCLEX PN Examination and succeed in your nursing career. NCLEX exam style multiple choice review questions at the end of each chapter test basic chapter knowledge as well as ask for appropriate nursing actions what the nurse should expect in terms of medical care of the patient and what complications the patient is at risk of developing. Case Studies and Applying Knowledge activities provide additional opportunities to apply learned information to clinical care. Thinking Critically activities require you to apply what you've learned in the textbook to new situations and draw conclusions based on that knowledge. Learning activities include matching labeling and completion exercises to help you learn basic concepts of maternity and pediatric nursing. Crossword puzzles in selected chapters are a nice way to change up the format when testing comprehension of terms and concepts. Answer key is located on the student's Evolve website. Several Thinking Critically sections and case studies with critical thinking questions have been added. Alternate item format questions are included in the review questions for the NCLEX Examination.

**Study Guide** Joseph J. Boyle, Douglas C. Giancoli, 1998 2000 2005 State Textbook Adoption Rowan Salisbury **Pamphlets on Protozoology (Kofoid Collection)**, 1912 **Study Guide** Kelly Bouas-Henry, 2005 By Kelly Bouas Henry Missouri Western State College and Douglas A Bernstein the Study Guide employs numerous techniques to help students learn. Each chapter including the optional Industrial Organizational Psychology chapter contains a detailed outline a key terms section that presents fresh examples and learning aids plus a fill in the blank test learning objectives a concepts and exercises section that shows students how to apply their knowledge of psychology to everyday issues and concerns a critical thinking exercise and personal learning activities. In addition each chapter concludes with a two part self quiz consisting of 40 multiple choice questions. An answer key tells the student not only which response is correct but also why each of the other choices is wrong and quiz analysis tables enable students to track patterns to their wrong answers either by topic or by type of question definition comprehension or application.

**Study Guide to Accompany Sociology** David Douglas Langlotz, Barbara D. Warne, 1996 **Study Guide for Lamanna and Riedmann's Marriages and Families** Paul O'Brien, 2000 **Study Guide Plus Sociology** BarBara Marliene Scott, 2000-09 **Study Guide to Accompany Raven and Johnson Biology** David A. Stetler, 1986 **Student Topic Outline and Study Guide to Accompany World of Chemistry, Essentials, Joesten/Netterville/Wood** James O. Schreck, 1993

**Telecourse Study Guide for Sociology** Pearson, James M. Henslin, Shelly Dutchin, 2005-04 This student manual is for students who are using Sociology A Down to Earth Approach with the twenty two video programs in the Exploring Society series from Dallas TeleLearning. Each section coordinates reading and video assignments and includes summaries learning

objectives video outlines key terms and people and student application projects The Guide also provides a self test section containing multiple choice true false fill in matching and essay questions that correlate to the video programs as well as Henslin s text **Prentice Hall Chemistry** ,2000

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