



# World Series In The Us

**Xiang Xie**



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web the six  $\text{NO}_3^-$  ions and the six  $\text{Na}^+$  ions that appear on both sides of the equation are spectator ions that can be canceled to give the net ionic equation  $3\text{Ba}^{2+} + 2\text{PO}_4^{3-} \rightarrow \text{Ba}_3(\text{PO}_4)_2\text{s}$

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web these spectator ions whose presence is required to maintain charge neutrality are neither chemically nor physically changed by the process and so they may be eliminated from the equation to yield a more succinct representation called a

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web video answer you know spectator ions are the ions present on the both sides of the equation indicate cancels out in the total ionic equations so you know in a case chemical reactions are described with chemical equations which symbolically present the starting materials end products

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web net ionic equations an equation for a reaction in solution showing only those particles that are directly involved in the chemical change spectator ions an ion that is not directly involved in a chemical reaction an ion that does not change oxidation number or composition during a reaction products

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web the equation below is the final net ionic equation after elimination of the spectator ions which yields water  $\text{H}^+ + \text{OH}^- \rightarrow \text{H}_2\text{O}$  the spectator ions in  $\text{KOH}$  and  $\text{HNO}_3$  are  $\text{K}^+$  and  $\text{NO}_3^-$  how to identify and list the

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reactions and products  $\text{pb no}_3$   $2 \text{ aq}$   $2 \text{kl aq}$   $\text{pb i}_2 \text{ s}$   $2 \text{kno}_3 \text{ aq}$  complete ionic equation an equation written with all soluble strong electrolytes shown as ions

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web the  $\text{ag aq}$  and  $\text{cl aq}$  ions become  $\text{agcl s}$  but the  $\text{na aq}$  ions and the  $\text{no}_3 \text{ aq}$  ions stay as  $\text{na aq}$  ions and  $\text{no}_3 \text{ aq}$  ions these two ions are examples of spectator ions ions that do nothing in the overall course of a chemical reaction

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web aug 8 2022 a spectator ion is an ion that does not take part in the chemical reaction and is found in solution both before and after the reaction the net ionic equation is the chemical equation that shows only those elements compounds and ions that are directly involved in the chemical reaction

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