

NAME: _____

STOICHIOMETRY WORKSHEET**A. Interpret the following reactions in terms of representative particles, moles and mass:**



B. Solve the following stoichiometry problems and write the correct answer with units.

1. How many moles of
- O_2
- can be produced by letting 12.00 moles of
- KClO_3
- react?



Answer: _____

2. How many moles of Cu are needed to react with 3.50 moles of
- AgNO_3
- ?



Answer: _____

3. How many moles of
- CO_2
- are produced with 10 mol of
- C_3H_8
- ?



Answer: _____

4. How many moles of HCl can be produced with 5.7 moles of
- H_2SO_4
- ?



Answer: _____

5. How many moles of Oxygen are needed to make complete combustion of 17.5 moles of propane,
- C_3H_8
- , in order to get
- CO_2
- and
- H_2O
- ?



Answer: _____

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