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you have o 2 16 2 32 grams mole 21 3 grams 0 66 moles 32 g m

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which reactant is limiting assuming we started with 30.0 grams of ammonium nitrate and 50.0 grams of sodium phosphate
what is the

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 i_2o_5 reacts with 28.0 grams of carbon monoxide c_o determine the mass of iodine i_2 which could be produced
80.0 g i_2o_5 1 mol i_2o_5 1 mol i_2 xs 1 333.8 g i_2o_5 1 mol i_2o_5 28 g c_o 1 mol c_o

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