

Answer Key

60 + 4 ext

Acids and Bases

How do acids and bases behave in water?

Why?

Acids and bases play an important role in our lives. Numerous biological processes, industrial applications, and even environmental problems are a function of the acidity or basicity (alkalinity) of aqueous solutions. It is therefore important to understand what makes a substance behave as an acid or a base when dissolved in water. In this activity, we will explore the physical and chemical properties of acids and bases.

Model 1 – Arrhenius Acids and Bases

Common Name for Aqueous Solution	Chemical Formula	Found in...	Tastes...	Turns Litmus Paper...	Conducts Electricity?	Acid or Base
Acetic acid	$\text{HC}_2\text{H}_3\text{O}_2(\text{aq})$	Vinegar	Sour	Red	Yes	Acid
Benzoic acid	$\text{HC}_7\text{H}_5\text{O}_2(\text{aq})$	Food preservative	Sour	Red	Yes	Acid
Phosphoric acid	$\text{H}_3\text{PO}_4(\text{aq})$	Soda pop	Sour	Red	Yes	Acid
Hydrochloric acid	$\text{HCl}(\text{aq})$	Stomach acid	Sour	Red	Yes	Acid
Citric acid	$\text{H}_3\text{C}_6\text{H}_5\text{O}_7(\text{aq})$	Citrus fruits	Sour	Red	Yes	Acid
Ascorbic acid	$\text{H}_2\text{C}_6\text{H}_7\text{O}_6(\text{aq})$	Vitamin C	Sour	Red	Yes	Acid
Magnesium hydroxide	$\text{Mg}(\text{OH})_2(\text{aq})$	Milk of magnesia	Bitter	Blue	Yes	Base
Aluminum hydroxide	$\text{Al}(\text{OH})_3(\text{aq})$	Antacids	Bitter	Blue	Yes	Base
Barium hydroxide	$\text{Ba}(\text{OH})_2(\text{aq})$	Lubricants	POISON	Blue	Yes	Base
Sodium hydroxide	$\text{NaOH}(\text{aq})$	Drain cleaner	POISON	Blue	Yes	Base

1. Refer to Model 1.
 - a. What is the common chemical name for Vitamin C?
Ascorbic acid
 - a. Is Vitamin C classified as an acid or a base?
Acid
2. Examine the properties of the Arrhenius acids in Model 1. List three properties that all Arrhenius acids have in common.
sour, turn litmus paper red, conduct electricity
3. Examine the chemical formulas for the Arrhenius acids in Model 1. What feature do all the Arrhenius acid chemical formulas have in common?
H at the beginning
4. Examine the properties of the Arrhenius bases in Model 1. List two properties that all Arrhenius bases have in common.
bitter / poison, turn litmus paper blue, conduct electricity

Suena Lab Manual Answer Key

John Barson

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Table of Contents Suena Lab Manual Answer Key

1. Understanding the eBook Suena Lab Manual Answer Key
 - The Rise of Digital Reading Suena Lab Manual Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Suena Lab Manual Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Suena Lab Manual Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Suena Lab Manual Answer Key
 - Personalized Recommendations
 - Suena Lab Manual Answer Key User Reviews and Ratings
 - Suena Lab Manual Answer Key and Bestseller Lists
5. Accessing Suena Lab Manual Answer Key Free and Paid eBooks
 - Suena Lab Manual Answer Key Public Domain eBooks
 - Suena Lab Manual Answer Key eBook Subscription Services
 - Suena Lab Manual Answer Key Budget-Friendly Options
6. Navigating Suena Lab Manual Answer Key eBook Formats

- ePub, PDF, MOBI, and More
- Suena Lab Manual Answer Key Compatibility with Devices
- Suena Lab Manual Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Suena Lab Manual Answer Key
 - Highlighting and Note-Taking Suena Lab Manual Answer Key
 - Interactive Elements Suena Lab Manual Answer Key
- 8. Staying Engaged with Suena Lab Manual Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Suena Lab Manual Answer Key
- 9. Balancing eBooks and Physical Books Suena Lab Manual Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Suena Lab Manual Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Suena Lab Manual Answer Key
 - Setting Reading Goals Suena Lab Manual Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Suena Lab Manual Answer Key
 - Fact-Checking eBook Content of Suena Lab Manual Answer Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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