

# Balancing Single Replacement Reactions

In a **single replacement** reaction, the cation (usually a metal) in a compound is replaced by another, more reactive metal. They always follow this format:



As an example, look at this chemical equation for single replacement reaction between solid lithium and a solution of iron (II) nitrate.



Lithium replaced iron (II) in the compound. Iron was left alone as a solid product. This reaction, however, is not balanced. Count the atoms. You might find it helpful to treat the nitrate polyatomic ion as a single element in balancing.

## Reactants

Li = \_\_\_\_\_

Fe = \_\_\_\_\_

NO<sub>3</sub> = \_\_\_\_\_

## Products

Li = \_\_\_\_\_

Fe = \_\_\_\_\_

NO<sub>3</sub> = \_\_\_\_\_

Add coefficients to balance the equation and satisfy the Law of Conservation of Matter.



Not all combinations of elements and compounds in single replacement reactions will actually react! You can use the **activity series** of metals to predict this. Follow these two rules:

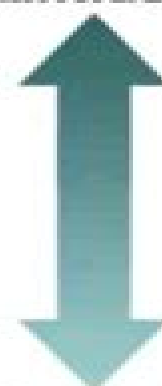
1. *Highly reactive metals prefer to be part of a compound.*
2. *Highly stable metals prefer to be alone.*

Look at the following combinations. If they will react, then predict what the products will be. If they will not react, then write NR for "no reaction".



lithium (Li)  
potassium (K)  
barium (Ba)  
calcium (Ca)  
sodium (Na)  
magnesium (Mg)  
aluminum (Al)  
zinc (Zn)  
cobalt (Co)  
nickel (Ni)  
lead (Pb)  
copper (Cu)  
silver (Ag)  
gold (Au)

most reactive



least reactive

# Single Replacement Reactions Study Guide

**ML Morrison**



## Single Replacement Reactions Study Guide:

**Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual** Leo J. Malone, Theodore O. Dolter, 2012-01-03  
The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section, and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts, and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem. Every concept in the text is clearly illustrated with one or more step-by-step examples. Making it Real essays have been updated to present timely and engaging real-world applications, emphasizing the relevance of the material they are learning. This edition continues the end-of-chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter. *Study Guide and Solutions Manual for Seager/Slabaugh's Chemistry for Today* Seager/Slabaugh, 2004. The fifth edition of the Study Guide and Student Solutions Manual has been updated to reflect all of the changes to the text. This ancillary tests the student on the learning objectives in each chapter and provides answers to all of the even-numbered end-of-chapter exercises. New additional activities have been added to include a review of each section of the chapter and a section entitled Tying It All Together with a Laboratory Application. ASVAB Study Guide Premium: 6 Practice Tests + Comprehensive Review + Online Practice Barron's Educational Series, Terry L. Duran, 2022-06-07. Be prepared for exam day with Barron's Trusted content from our experts. Barron's ASVAB Study Guide Premium includes everything you need to be prepared for exam day with comprehensive review and practice from an experienced ASVAB expert. All the Review You Need to Be Prepared. An expert overview of the ASVAB. In-depth subject review covering all sections of the test. Tips and strategies from Barron's expert author. Practice with Confidence. 6 full-length practice tests. 3 in the book and 3 online, including 1 diagnostic test and 1 AFQT-focused assessment. Review chapters contain additional practice questions. All practice questions include detailed answer explanations. Interactive Online Practice. 3 full-length practice tests online with a timed test option to simulate exam experience. AFQT-focused option for each test. Detailed answer explanations included with expert advice. Automated scoring to check your learning progress. *Hazmat Chemistry Study Guide (Second Edition)* Jill Meryl Levy, 2011. **Student Study Guide and Solutions Manual to Accompany General, Organic, and Biochemistry** Katherine J. Denniston, Denniston Katherine, 2006. **Study Guide** Darrell D. Ebbing, Susan M. Schelble, 1998. Basic Concepts of Chemistry, Study Guide Leo J. Malone, 2000. The text's three main goals are to introduce chemistry as a living, relevant science to encourage learning and critical thinking and to help readers overcome the math difficulties that impede their progress in chemistry. Designed to help readers master the principles of

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