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the savings What is Cyber Monday and will it ever be Black Friday What is Travel Tuesday How do I stay safe online Where are the best deals What are some of the latest scams Is Black Friday worth it Is it better to shop online or in the store What goods should I buy in the store What should I buy online How do I keep my bank account secure What websites should I visit to help me find deals What kind of mindset do I need to become a savvy shopper Should I buy electronics online About The Expert Tammy Marshall is a savvy shopper She s been one her entire life She lives to save money Her other passion is sharing how she finds new ways to save green Tammy also deplores cyber crooks Her mission is to uncover the latest scams so she can share them with you and help you protect your wallets She s a professional journalist and has been writing for 20 years When she isn t finding new ways to spend less she is working on her family farm and growing her own veggies and displaying them at the county fair She s loves growing food like she loves to save dollars For her she doesn t just enjoy the thrill of shopping She enjoys seeing prices drop while she fills her cart and her pantry at home Here s my video <https://www.youtube.com/watch?v=7N5yxqdUF5Q> HowExpert publishes short how to guides on unique topics by everyday experts

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**Black Friday and Cyber Monday Guide**

HowExpert,Marshall Tammy,2019-11-09 Have you ever wondered how to save during the holidays Presents and pampering guests there are all sorts of ways that you willingly spend during this season The trick you may not know is that there are special days during the holiday season that you will save the most money Stores know this you can too Our guide is designed to help you save your green during the festivities You won t be a scrooge when you get your hands on the savings we ve listed In this book you will find answers to the following questions What is Black Friday and how do I get in on the savings What is Cyber Monday and will it ever be Black Friday What is Travel Tuesday How do I stay safe online Where are the best deals What are some of the latest scams Is Black Friday worth it Is it better to shop online or in the store What goods should I buy in the store What should I buy online How do I keep my bank account secure What websites should I visit to help me find deals What kind of mindset do I need to become a savvy shopper Should I buy electronics online About The Expert Tammy Marshall is a savvy shopper She s been one her entire life She lives to save money Her other passion is sharing how she finds new ways to save green Tammy also deplores cyber crooks Her mission is to uncover the latest scams so she can share them with you and help you protect your wallets She s a professional journalist and has been writing for 20 years When she isn t

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electrons each with the same spin an atomic orbital can hold a minimum of 2 electrons each with opposite spins multiple choice edit please save your changes before editing any questions

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web the electronic configuration of an atom in the quantum mechanical model is stated by listing the occupied orbitals in order of filling with the number of electrons in each orbital indicated by superscript in this notation the electronic configuration of sodium would be  $1s^2 2s^2 2p^6 3s^1$  distributed in the orbitals as

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web the orbital diagram of helium is  $1s^2$  the spin of an electron is specified by a fourth quantum number called the the two possible values of  $m_s$  are spin up and spin down the three principles that are needed to specify orbital occupancy for the electrons are electrons and they must have pauli exclusion principle no orbital may have more than

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web the electron configuration and the orbital diagram are following hydrogen is the noble gas helium which has an atomic number of 2 the helium atom contains two protons and two electrons the first electron has the same four quantum numbers as the hydrogen atom electron  $n=1, l=0, m_l=0, m_s=1/2$

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web determine element from electron configuration determine number of electrons in an element s orbital s element groups to use select at least one s block alkali metals and alkaline earth metals p block nonmetals metalloids and other metals d

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