

# Balancing Word and Skeleton Equations

Write and balance the skeleton equation for each of the following reactions.

- a** calcium carbide + oxygen  $\rightarrow$  calcium + carbon dioxide
  
- b** chromium + tin (IV) chloride  $\rightarrow$  chromium (III) chloride + tin
  
- c** potassium hydroxide + hydrogen bromide  $\rightarrow$  potassium bromide + water
  
- d** aluminum oxide  $\rightarrow$  aluminum + oxygen gas
  
- e** propane + oxygen  $\rightarrow$  carbon dioxide + water
  
- f** barium carbonate  $\rightarrow$  barium oxide + carbon dioxide
  
- g** iron + sulfur  $\rightarrow$  iron (II) sulfide
  
- h** aluminum + copper (II) chloride  $\rightarrow$  aluminum chloride + copper
  
- i** silver nitrate + copper  $\rightarrow$  silver + copper nitrate
  
- j** hydrogen + chlorine  $\rightarrow$  hydrogen chloride

# Word And Skeleton Equations Answers

**Alexander Smith**

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