

Name _____ Date _____ Class _____

CHAPTER 3 STUDY GUIDE FOR CONTENT MASTERY

Matter—Properties and Changes**Section 3.1 Properties of Matter**

Fill in the blank, read about physical properties and chemical properties of matter.

Use each of the terms below just once to complete the passage.

chemical density	mass properties	physical substances
---------------------	--------------------	------------------------

Matter is anything with **mass** and **volume**.A **substance** is a form of matter with a constant and unchanging composition.Substances have specific, unchanged **properties** that can be observed.Substances have both **physical** and **chemical** properties. **Physical**properties can be observed without changing a substance's chemical composition. Color, hardness, and **density**, for example. Other properties cannot be observed without changing the composition of a substance. These are called**chemical** properties. An example is the tendency of zinc to form rust when exposed to air.Label each property as either **physical** or **chemical**.**physical** 1. Chemical formula H_2O **chemical** 2. Rusts green substance when exposed to air**chemical** 3. Remains unchanged when in the presence of nitrogen**physical** 4. Colorless**physical** 5. Solid at normal temperatures and pressures**chemical** 6. Ability to combine with another substance**physical** 7. Melting point**physical** 8. Liquid at normal temperatures and pressures**physical** 9. Boiling point is 100°C **physical** 10. Conducts electricity**physical** 11. Density is $\frac{10}{10}$

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CHAPTER 3 STUDY GUIDE FOR CONTENT MASTERY

Section 3.1 continued

In your textbook, read about states of matter.

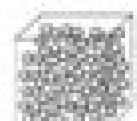
Label each drawing with one of these words: solid, liquid, gas.



gas



solid



liquid

For each statement below, write true or false.

True _____ 11. All matter in everyday life exists in one of these physical forms.

True _____ 12. A solid has definite shape and volume.

False _____ 13. A liquid has a definite shape and takes on the volume of its container.

True _____ 14. A gas has both the shape and the volume of its container.

False _____ 15. The particles in a gas cannot be compressed into a smaller volume.

False _____ 16. Liquids tend to contract when heated.

False _____ 17. The particles in a solid are spaced far apart.

False _____ 18. The words gas and vapor can be used interchangeably.

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Promotion Fitness Examination Study Guide ,2003 Study Guide with Answers to Selected Problems H. Stephen Stoker,Danny V. White,2003-04 This supplement includes for each chapter a brief overview activities and practice problems to reinforce skills and a practice test The answers section includes answers for all odd numbered end of chapter exercises

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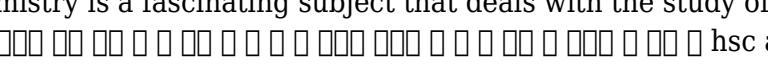
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