

Study Guide Unit 9

1. KNOW the following vocabulary words (this means you NEED to be able to DEFINE them ALL):

- a. Solution-
m. Solute-
- b. Homogeneous-
n. Solvent-
- c. Heterogeneous-
- d. Soluble-
o. Acid-
- e. Insoluble-
p. Base-
- f. Miscible-
q. Titration-
- g. Immiscible-
r. Analyte-
- h. Solubility-
s. Titrant-
- i. Alloy-
t. Monoprotic acid-
- j. Saturated-
u. Diprotic Acid
- k. Unsaturated-
v. Triprotic Acid
- l. Supersaturated-

2. How is the solubility of a gas affected by temp. and pressure?

Pressure up = solubility up

Temp up = solubility down

3. Under what conditions are salts most soluble?

High temperatures

4. Use the solubility curve to the right to answer the following.

a. What mass of KNO_3 will make a saturated solution at 60°C ? **100g**

b. Which of the compounds shown on the solubility curve decrease solubility as temperature increases?

NH_3 , $\text{Ca}_2(\text{SO}_4)_3$

c. What type of solution do I have if I **add** 110 g of NaNO_3 in 100g of 20°C water?

Saturated

d. What mass of KCl can go into solution at 80°C ?

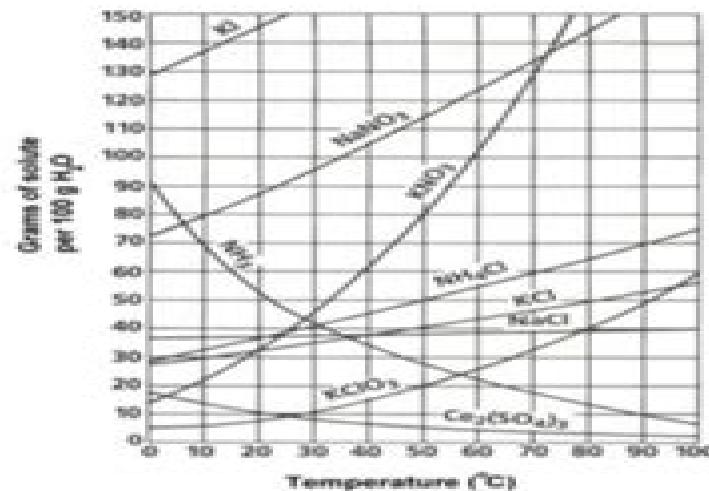
50 g

e. What is the least soluble compound at 20°C ?

KClO_3

f. Excluding KI , what is the most soluble compound at 80°C ?

KNO_3



Study Guide Solutions Answer Key

Irvin N. Gleim

Study Guide Solutions Answer Key:

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Table of Contents Study Guide Solutions Answer Key

1. Understanding the eBook Study Guide Solutions Answer Key
 - The Rise of Digital Reading Study Guide Solutions Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Study Guide Solutions Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Study Guide Solutions Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Study Guide Solutions Answer Key

- Personalized Recommendations
- Study Guide Solutions Answer Key User Reviews and Ratings
- Study Guide Solutions Answer Key and Bestseller Lists

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- Study Guide Solutions Answer Key Budget-Friendly Options

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- ePUB, PDF, MOBI, and More
- Study Guide Solutions Answer Key Compatibility with Devices
- Study Guide Solutions Answer Key Enhanced eBook Features

7. Enhancing Your Reading Experience

- Adjustable Fonts and Text Sizes of Study Guide Solutions Answer Key
- Highlighting and Note-Taking Study Guide Solutions Answer Key
- Interactive Elements Study Guide Solutions Answer Key

8. Staying Engaged with Study Guide Solutions Answer Key

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Study Guide Solutions Answer Key

9. Balancing eBooks and Physical Books Study Guide Solutions Answer Key

- Benefits of a Digital Library
- Creating a Diverse Reading Collection Study Guide Solutions Answer Key

10. Overcoming Reading Challenges

- Dealing with Digital Eye Strain
- Minimizing Distractions
- Managing Screen Time

11. Cultivating a Reading Routine Study Guide Solutions Answer Key

- Setting Reading Goals Study Guide Solutions Answer Key
- Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Study Guide Solutions Answer Key

- Fact-Checking eBook Content of Study Guide Solutions Answer Key
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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