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for the nitration 4 01 ml of nitric acid 4 00 ml of sulfuric acid and 4 56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4 56 ml of bromobenzene was used the amount of product predicted to be recovered was 8 7533 g

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the aim of the experiment is to determine and study electrophilic aromatic substitution reaction eas particularly nitration of methyl benzoate go give methyl m nitrobenzoate thus determine the weight melting point and percentage yield of the pure methyl m nitrobenzoate abstract

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in this experiment we will carry out the nitration of bromobenzene bromine substituted on an aromatic ring is an ortho para directing group it will direct an incoming electrophile to the ortho and para positions it is also deactivating meaning that its reactions will be slower than the identical reactions of benzene

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section 20711 lab experiment nitration of bromobenzene introduction the chemical reaction of nitration consists of a nitro group being added to or substituted in a molecule nitration can basically be carried out by a mixture of concentrated nitric acid and sulfuric acid this mixture is useful to obtain the active nitronium ion

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1 go to the restroom and wash the affected area with lots of soap and water the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area 3 the summary to this warning is not to touch any part of your body while performing this experiment if you must leave the lab for any reason 1

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in the nitration of bromobenzene a mixture of sulfuric acid and nitric acid react with one another nitric acid the weaker of the two acids will serve as base to form nitronium ion bromobenzene will attack the nitronium ion by nucleophilic attack

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1 bromo 4 nitrobenzene

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture HNO_3 3 h 2 so 4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0.6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accomodate their electrons demonstration of creating the p arenium cation by to nitration of bromobenzene

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion NO_2^+ which acts as the electrophilic species

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neue regeln für minijobs das ändert sich merkur de - Jul 10 2022

jan 30 2021 a minijob esetén csak maximálisan 520 eurót kereshetsz és nem vonnak le adót vagy más járulékokat a midijob ról itt írtunk részletesen kevés vagy nulla nyelvtudást

minijob definition und regelungen sage - May 08 2022

sep 25 2022 zum 1 oktober steigt die minijobgrenze von 450 auf 520 euro nach wie vor gilt die jobs sind für arbeitnehmer in der regel steuer und sozialversicherungsfrei ausnahme

minijob grenze 2022 und 2023 personal haufe - Jul 22 2023

eine geringfügige beschäftigung liegt demnach vor wenn erstens das arbeitsentgelt aus dieser beschäftigung regelmäßig im monat 450 euro nicht übersteigt oder zweitens die

minijob rechte und pflichten im 450 euro job devk - Dec 15 2022

durch fehlende beiträge zu den sozialversicherungen sichern minijobs sozial nicht ab auch ehemals 450 euro job 520 euro job geringfügige beschäftigung kurzfristige