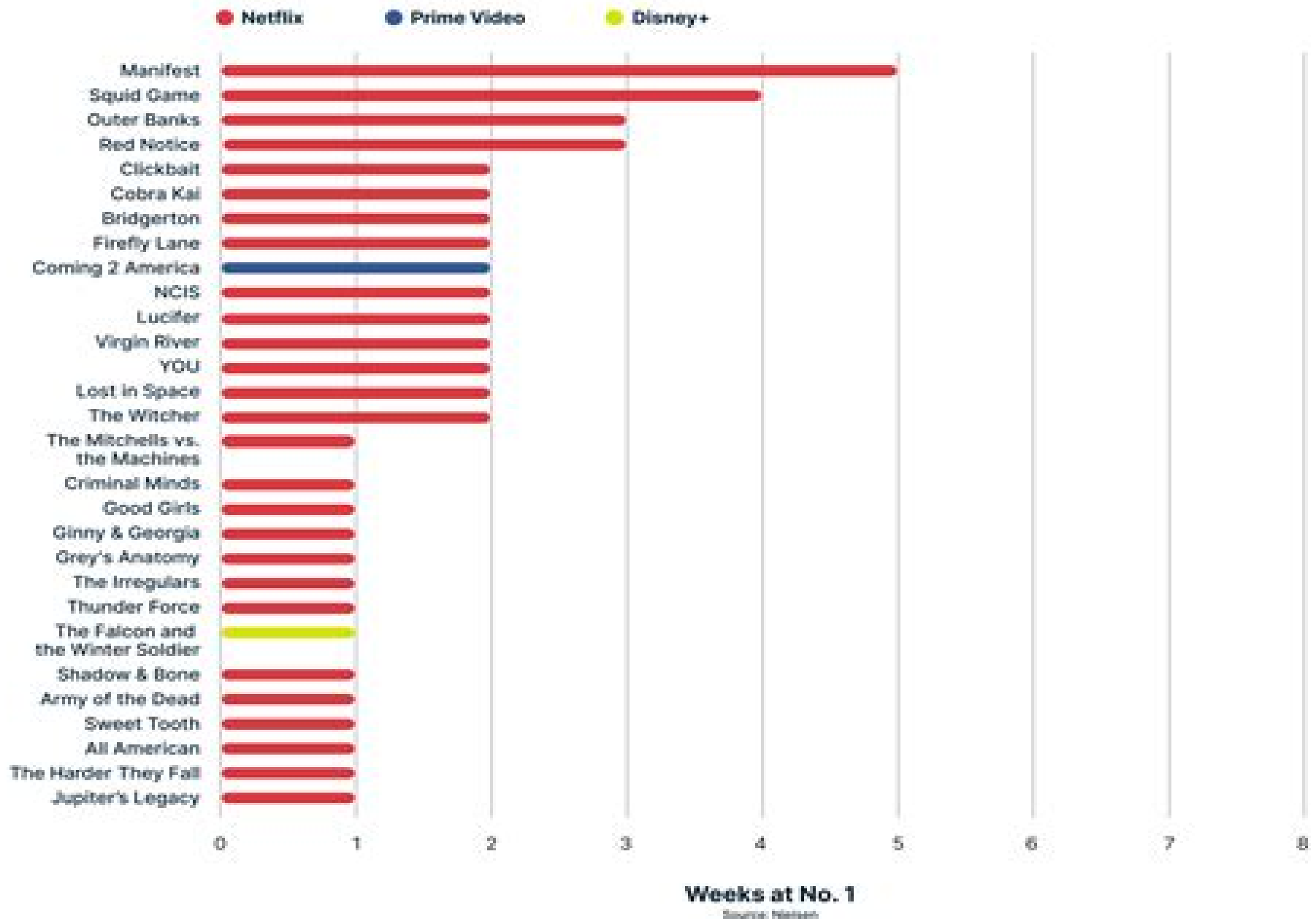


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 $3\text{Mg} + 1\text{Fe}_2\text{O}_3 \rightarrow 2\text{Fe} + 3\text{MgO}$ type of reaction single displacement
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 $\text{CH}_2\text{O} + \text{CO}_2 + \text{H}_2\text{O} \rightarrow \text{C}_2\text{H}_4 + 2\text{O}_2$
 $2\text{Fe} + 6\text{NaBr} \rightarrow 2\text{FeBr}_3 + 6\text{Na}$
 $\text{CaSO}_4 + \text{MgSO}_4 + 2\text{H}_2\text{O} + 4\text{NH}_4\text{OH} \rightarrow \text{H}_2\text{O} + \text{NH}_4\text{Br} + \text{PbO}_2 + \text{Na}_2\text{CO}_3 + \text{Na}_2\text{O} + \text{CO}_2 + \text{C}_4\text{H}_8 + 6\text{O}_4 + 2\text{H}_2\text{O} + \text{HCl} + \text{NaOH} + \text{H}_2\text{O} + \text{NaCl}$

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