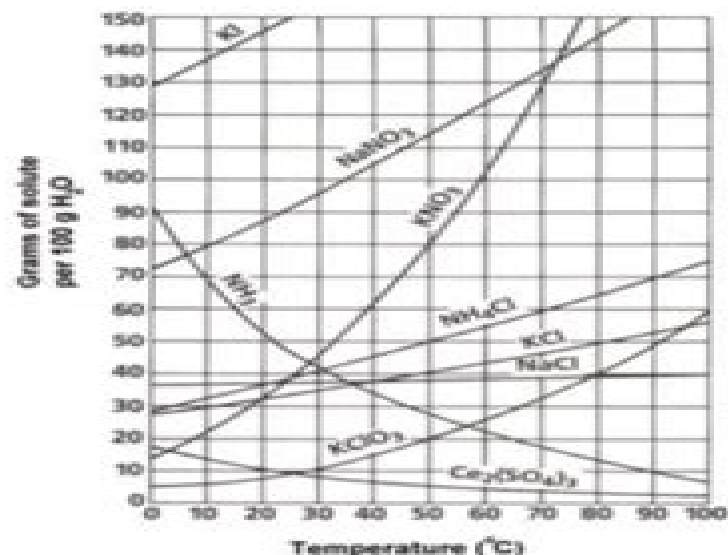


Chemistry I-H
Solubility Curves and Solutions Review Sheet
Solution Set

1. Define solubility: **The maximum amount of solute that dissolves in a fixed quantity of a particular solvent at a specified temperature.**

Use the solubility curve below to answer questions #2 - #15. Be sure to note the units on the axes of the graph.



2. In general, how does temperature affect solubility?
The higher the temp. the more substance dissolves
3. **KClO₃** Which compound is least soluble at 10°C?
4. **50 g** How many grams of KCl can be dissolved in 100g of water at 80°C?
5. **40 g** How many grams of NaCl can be dissolved in 100g of water at 90°C?
6. **about 60 g** At 40°C, how much KNO₃ can be dissolved in 100g of water?
7. **NaCl** Which compound shows the least amount of change in solubility from 0°C-100°C?
8. **unsaturated** At 30°C, 90g of NaNO₃ is dissolved in 100g of water. Is this solution saturated or unsaturated?
9. **D** At 60 C, 72 g of NH₄Cl are dissolved in 100g of water. This solution is considered to be:
A) saturated; B) unsaturated; C) supersaturated; D) saturated with some left undissolved
10. **30 g** A saturated solution of KClO₃ is formed from one hundred grams of water. If the saturated solution is cooled from 90°C to 50°C, how many grams of precipitate are formed?
11. **20 g** A saturated solution of NH₄Cl is formed from one hundred grams of water. If the saturated solution is cooled from 80°C to 40°C, how many grams of precipitate are formed?
12. **E** Which compounds show a *decrease* in solubility from 0°C-100°C?
A) potassium chlorate B) cerium sulfate C) sodium nitrate D) ammonia E) Both B and D
13. **C** Which compound is the most soluble at 10°C?
A) Ce₂(SO₄)₃ B) NaCl C) KCl D) KClO₃ E) NH₃
14. **C** A chemist adds 120 grams of KCl into a test tube which contains 100 grams of water. The temperature of the water is 80°C. How much of the solute will be left undissolved at the bottom of the test tube?
A) 50 grams B) 60 grams C) 70 grams D) 80 grams E) 56 grams

Solutions And Solubility Review Sheet

TD Snyder



Solutions And Solubility Review Sheet:

Chemistry and Industry Review ,1929 **New York Review of the Telegraph and Telephone and Electrical Journal** ,1902 **The Electrical Review** ,1890 *Electrical Review* ,1902 **Toxicity of Nanomaterials** Suresh Pillai,Yvonne Lang,2019-05-16 Choice Recommended Title April 2020 This comprehensive book edited by two leading experts in nanotechnology and bioengineering with contributions from a global team of specialists provides a detailed overview of the environmental and health impacts associated with the toxicology of nanomaterials Special attention is given to nanomaterial toxicity during synthesis production and application and chapters throughout are focused on key areas that are important for future research and development of nanomaterials This book will be of interest to advanced students studying biomedical engineering and materials science PhD researchers post docs and academics working in the area of nanotechnology medicine manufacturing and regulatory bodies Features Collates and critically evaluates various aspects of the toxicology of nanomaterials in one comprehensive text Discusses the various effects of nanocrystals including the morphologies on cytotoxicity in addition to the environmental and cytotoxicity risks of graphene and 2D nanomaterials Explores practical methods of detection and quantification with applications in the environmental and healthcare fields *India Rubber World and Electrical Trades Review* John Robertson Dunlap, Henry Clemens Pearson,1921 Keywords Index to U.S. Government Technical Reports United States. Department of Commerce. Office of Technical Services,1962 **The Laboratory Study of Chemistry** Herbert Royle Smith,1918 *Pharmaceutical Record and Weekly Market Review* P. W. Bedford,1890 *The Technical Review* ,1920 **Platinum Metals Review** ,1978 **Technical Association of the Pulp and Paper Industry** ,1963 **The Chemical News and Journal of Industrial Science** ,1913 *The Chemical News and Journal of Physical Science* ,1913 **Iron & Coal Trades Review** ,1927 Vol 115 includes Diamond jubilee issue 1867 1927 *Electro-chemist and Metallurgist and Metallurgican Review* ,1901 **The Journal of the Engineering Institute of Canada** Engineering Institute of Canada,1919 **Chemical News and Journal of Physical Science** ,1875 **Chemical News and Journal of Industrial Science** ,1905 **The Journal of the Chemical, Metallurgical & Mining Society of South Africa** ,1923

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