

6. What type of bond(s) were formed in the compound from question #4? *Ionic Bond*

7. Draw the bond(s) that will form between P and Cl.



8. What is the formula for the compound formed in question #7?



9. What type of bond(s) were formed in the compound from question #7? *Covalent Bond*

10. Draw the correct valence electron configuration for each atom given.

H	<i>1</i>	<i>Ca</i>	<i>Al</i>
<i>Ne</i>	<i>Ge</i>	<i>B</i>	<i>K</i>
<i>Li</i>	<i>Si</i>	<i>C</i>	<i>Xe</i>

11. Fill in the table with the correct values.

8 - Valence e- (except H)

Element	# valence electrons	# covalent bonds able to be formed
H	1	1
He	2	0
C	4	4
Cl	7	1
Ar	8	0
O	6	2
N	5	3
P	5	3

Study Guide Answer Key For Chemical Bond

Leo J. Malone

Study Guide Answer Key For Chemical Bond:

Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03

The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem. Every concept in the text is clearly illustrated with one or more step by step examples. Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning. This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter.

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