

CHAPTER 3 STUDY GUIDE FOR CONTENT MASTERY

Matter—Properties and Changes**Section 3.1 Properties of Matter**

Fill in the blank, read about physical properties and chemical properties of matter.

Use each of the terms below just once to complete the passage.

chemical density	mass properties	physical substances
---------------------	--------------------	------------------------

Matter is anything with **mass** and **volume**. A **substance** is a form of matter with a uniform and unchanging composition. Substances have specific, unchanged **properties** that can be observed. Substances have both **physical** and **chemical** properties. **Physical** properties can be observed without changing a substance's chemical composition. Color, hardness, and **density** are examples. Other properties cannot be observed without changing the composition of a substance. These are called **chemical** properties. An example is the tendency of zinc to form rust when exposed to air.

Label each property as either physical or chemical.

physical 3. Chemical formula H_2O

chemical 4. Rusts green substance when exposed to air

chemical 5. Remains unchanged when in the presence of nitrogen

physical 6. Colorless

physical 7. Solid at normal temperatures and pressures

chemical 8. Ability to combine with another substance

physical 9. Melting point

physical 10. Liquid at normal temperatures and pressures

physical 11. Boiling point is 100°C

physical 12. Conducts electricity

physical 13. Density is $\frac{1}{20}$

CHAPTER 3

STUDY GUIDE FOR CONTENT MASTERY

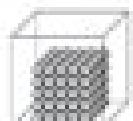
Section 3.1 continued

In your textbook, read about states of matter.

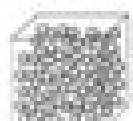
Label each drawing with one of these words: solid, liquid, gas.



gas



solid



liquid

For each statement below, write true or false.

true 15. All matter in everyday life exists in one of these physical forms.

true 16. A solid has definite shape and volume.

false 17. A liquid has a definite shape and takes on the volume of its container.

true 18. A gas has both the shape and the volume of its container.

false 19. The particles in a gas cannot be compressed into a smaller volume.

false 20. Liquids tend to contract when heated.

false 21. The particles in a solid are spaced far apart.

false 22. The words gas and vapor can be used interchangeably.

Study Guide Matter And Change Key

Karen Timberlakes

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Promotion Fitness Examination Study Guide,2003 *Study Guide with Answers to Selected Problems* H. Stephen Stoker,Danny V. White,2003-04 This supplement includes for each chapter a brief overview activities and practice problems to reinforce skills and a practice test The answers section includes answers for all odd numbered end of chapter exercises

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web this chapter is intended to familiarize you with how maple works how expressions and commands are entered in maple

and how to make sense of the results