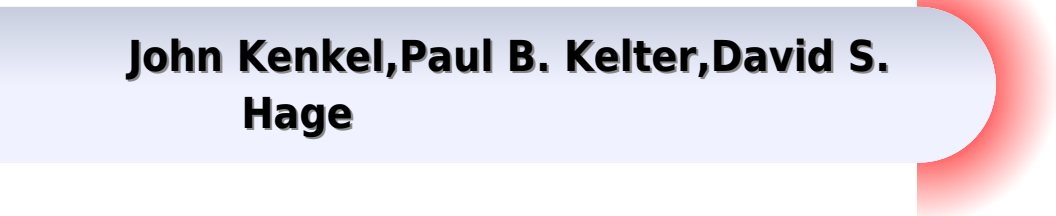


## NUCLEAR CHEMISTRY STUDY GUIDE—CH. 21

1. Write the symbol for uranium—238 in isotope format (with top and bottom #s).
2. What does the top # represent?
3. What does the bottom # represent?
4. What particles are found in the nucleus?
5. Why does the strong nuclear force exist?
6. What is the electrostatic force?
7. What particles does the electrostatic force deal with?
8. Define nuclear reactions.
9. Define nuclear fission and fusion.
10. Nuclear reactions usually produce \_\_\_\_\_ amounts of energy, and isotopes become \_\_\_\_\_ stable. The \_\_\_\_\_ number and the \_\_\_\_\_ number may also change.
11. What is transmutation?
12. Write the beta decay of radium—226.
13. Write the alpha decay of radium—226.
14. Why would the isotopes and the end of problems #13 and #14 still be unstable?
15. Define nuclear radiation.
16. List the 3 radioactive particles in order of INCREASING penetrating power.
17. What is radioactive decay?
18. Alpha particles are basically \_\_\_\_\_ nuclei, and beta particles are basically high energy \_\_\_\_\_.
19. What is half life?
20. What is radioactive dating?
21. What is cobalt—60 used for?
22. Does fission or fusion produce nuclei with heavier mass than the original mass?
23. Name 3 differences between chemical and nuclear reactions.
24. What is carbon—13 radioactive?
25. Elements past atomic # \_\_\_\_\_ are radioactive.
26. If  $1/16$  of a sample of a radioactive isotope is left after nuclear decay, how long did this take if the  $\frac{1}{2}$  life is 4 days?
27. If the half life of a radioactive isotope is 4 days, how much of a 50 gram sample is left after 12 days?

# Study Guide Nuclear Chemistry

**John Kenkel, Paul B. Kelter, David S.  
Hage**



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**Radiochemistry and Nuclear Chemistry** Gregory Choppin, Jan Rydberg, Jan-Olov Liljenzin, 2001-12-03

Radiochemistry or Nuclear Chemistry is the study of radiation from an atomic or molecular perspective including elemental transformation and reaction effects as well as physical health and medical properties This revised edition of one of the earliest and best known books on the subject has been updated to bring into teaching the latest developments in research and the current hot topics in the field In order to further enhance the functionality of this text the authors have added numerous teaching aids that include an interactive website that features testing examples in MathCAD with variable quantities and options hotlinks to relevant text sections from the book and online self grading texts As in the previous edition readers can closely follow the structure of the chapters from the broad introduction through the more in depth descriptions of radiochemistry then nuclear radiation chemistry and finally the guide to nuclear energy including energy production fuel cycle and waste management New edition of a well known respected text in the specialized field of nuclear radiochemistry Includes an interactive website with testing and evaluation modules based on exercises in the book Suitable for both radiochemistry and nuclear chemistry courses

**Chemistry, Student Study Guide** John A. Olmsted, Gregory M.

Williams, 2002-01-04 This third edition continues to innovate by providing students with an integrated and modern approach to the subject The text emphasizes the modern tools of chemistry while incorporating historical evidence and its unique molecular quantitative emphasis is further reinforced by an integrated media package developed by the authors Also of benefit is the just in time presentation of key content only providing details once they are needed While key topics and

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Robinson Chemistry Study Guide Robinson,1992      Students Study Guide Fredeen,Pearson,2003-07      **Study Guide for General Chemistry and College Chemistry, Eighth Editions by Holtzclaw and Robinson** Norman E. Griswold, Henry Fuller Holtzclaw,1988      **Student Study Guide/Solutions Manual for Essentials of General, Organic, and Biochemistry** Denise Guinn, Rebecca Brewer, Rachel C. Lum,2009-09-15 The Student Study Guide and Solutions Manual provides students with a combined manual designed to help them avoid common mistakes and understand key concepts After a brief review of each section s critical ideas students are taken through stepped out worked examples try it yourself examples and chapter quizzes all structured to reinforce chapter objectives and build problem solving techniques The solutions manual includes detailed solutions to all odd numbered exercises in the text      **Chemistry Mark** Jackson,2012-05-31 BarCharts best selling quick reference to chemistry has been updated and expanded in this new edition With updated content and an additional panel of information this popular guide is not only an essential companion for students in introductory chemistry courses but also a must have refresher for students in higher level courses Author Mark D Jackson PhD a scientist and university chemistry professor has a gift for making the complicated subject of chemistry interesting and easy to understand without the fluff In this new edition you will find more coverage of the subject helpful illustrations chemical problems and practical applications making this a study tool you won t want to be without

*Introduction to Chemistry, Study Guide* T. R. Dickson,1999-07-07 Teaches chemistry by offering a dynamic provocative and relevant view of the topic and its importance to society and our daily lives Three themes are stressed throughout the text developing chemical thinking and chemical vision and refining problem solving skills Many chapters in this edition has been rewritten and rearranged to vitalize the topics and to include interesting examples analogies and images      Student Study Guide to Accompany Petrucci's General Chemistry, 3rd. Ed Robert K. Wismer,1982      *Study Guide to Accompany Chemical Principles, Properties, and Reactions* Kenneth L. Henold,1984      **Chemistry** John Kenkel, Paul B. Kelter, David S. Hage,2000-09-21 What a great idea an introductory chemistry text that connects students to the workplace of practicing chemists and chemical technicians Tying chemistry fundamentals to the reality of industrial life Chemistry An Industry Based Introduction with CD ROM covers all the basic principles of chemistry including formulas and names chemical bon      **CLEP Official Study Guide** College Entrance Examination Board,2006-08 A guide to preparing for the thirty four different kinds of college CLEP exams that includes information on deciding which exams to take interpreting the scores taking the tests and other related topics      *Study Guide for Chemical Principles* Thomas Elliott Taylor,1979      **Essentials of Nuclear**

**Chemistry** Hari Jeevan Arnikar, 1987-04-15 A thorough introduction to the essential topics of nuclear chemistry With clarity and illustrative examples it covers nuclear structure and stability types of radioactivity and nuclear reactions and the processes of nuclear fission and fusion This edition offers clearer and more up to date coverage of the subject and incorporates entirely new material as well New to this edition a detailed account of nuclear magnetic resonance coverage of the differences and limitations of the Gamov Teller and Fermi selection rules and examples of the earliest nuclear reactions in the cosmos Special attention is paid to the study of magnetic moments of elementary particles and nuclei Features numerical examples with answers and a unique and helpful inclusion of historically important and interesting events

**Chemical Principles** Peter Atkins, Loretta Jones, 2010 This text is designed for a rigorous course in introductory chemistry Its central theme is to challenge students to think and question while providing a sound foundation in the principles of chemistry *College Chemistry* Steven Boone, Drew H. Wolfe, 2011-11-01 The Collins College Outline for College Chemistry is a comprehensive guide to the fundamental concepts behind chemical reactions bonding equilibria and thermodynamics with topics ranging from simple chemical measurements and the basics of atoms and molecules to entropy electrochemistry and nuclear chemistry Fully revised and updated by Dr Steven Boone College Chemistry includes practical test yourself sections with answers and complete explanations at the end of each chapter Also included are essential vocabulary definitions and sample exercises as well as detailed images charts and diagrams The Collins College Outlines are a completely revised in depth series of study guides for all areas of study including the Humanities Social Sciences Mathematics Science Language History and Business Featuring the most up to date information each book is written by a seasoned professor in the field and focuses on a simplified and general overview of the subject for college students and where appropriate Advanced Placement students Each Collins College Outline is fully integrated with the major curriculum for its subject and is a perfect supplement for any standard textbook **CliffsAP Chemistry, 4th Edition** Bobrow Test Preparation Services, 2011-09-26 Your complete guide to a higher score on the AP Chemistry exam Why CliffsAP Guides Go with the name you know and trust Get the information you need fast Written by test prep specialists Contents include Introduction overview of the test and how it is scored proven strategies for each type of question Review of topics tested atom periodic table bonding geometry hybridization stoichiometry gases liquids and solids thermodynamics solutions equilibrium acids and bases kinetics redox nuclear chemistry organic chemistry and writing reactions The Labs feature 20 multiple choice questions multiple free response questions on each topic with answers on each topic with answers and explanations scoring rubrics and 2 full length practice exams Structured like the actual exam Complete with answers and explanations AP is a registered trademark of the College Board which was not involved in the production of and does not endorse this product NES Chemistry Study Guide NES Chemistry Exam Prep Team, 2018-02-12 Imagine a study guide actually designed for teachers Because we know you've got a busy life we've developed a study guide that isn't like other

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**Nuclear Chemistry** Maheshwar Sharon, Madhuri Sharon, 2021-02-01 This book is designed to serve as a textbook for core courses offered to postgraduate students enrolled in chemistry This book can also be used as a core or supplementary text for nuclear chemistry courses offered to students of chemical engineering The book covers various topics of nuclear chemistry like Shell model fission fusion reaction natural radioactive equilibrium series nuclear reactions carried by various types of accelerators In addition it describes the law of decay of radioactivity type of decay and interaction of radiation with matter It explains the difference between ionization counter scintillation counter and solid state detector This book also consists of end of book problems to help readers aid self learning The detailed coverage and pedagogical tools make this an ideal textbook for postgraduate students and researchers enrolled in various chemistry and engineering courses This book will also be beneficial for industry professionals in the allied fields

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