

# 1. Light and Quantum Energy: The Wave Nature of Light

## The Photoelectric Effect (Einstein's Model)

- 1. Did not contradict the wave's characteristics arranged under special circumstances
- 2. Did not address the question of why the negative's electron ejection was not gradual over time
- 3. Did not conclude that the distribution and distribution of electron ejection among the various elements

## Wave Properties

- 1. Wave properties of light are characterized by:
- 2. Frequency: number of waves that pass a point per second
- 3. Wavelength: frequency of the light waves is the amplitude of the wave
- 4. The wave's energy is proportional to the square of the amplitude
- 5. The wave's energy is proportional to the square of the amplitude
- 6. The wave's energy is proportional to the square of the amplitude



## Wave Properties

- 1. All waves travel at the same speed ( $c = 3.00 \times 10^8$  m/s)
- 2. The frequency of a wave is the number of waves that pass a point per second
- 3. The wavelength of a wave is the distance between two consecutive peaks
- 4. The speed of light is the product of the frequency and the wavelength
- 5. The frequency of a wave is the number of waves that pass a point per second
- 6. The wavelength of a wave is the distance between two consecutive peaks

## Wave Properties: Wavelength



# Study Guide Light And Quantized Energy

**Martha B. Barrett**



## Study Guide Light And Quantized Energy:

**Chemistry, Student Study Guide** James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane  $\text{CH}_4$  from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement

**Physics for Scientists and Engineers Study Guide** Paul A. Tipler, Gene Mosca, 2008-01-11 The study guide provides students with key physical quantities and equations misconceptions to avoid questions and practice problems to gain further understanding of physics concepts and quizzes to test student knowledge of chapters All written with the same level of detail as the examples found in the text

Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter A new Math Check allows quick access to the needed basic skill The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter

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