

# The Mole Study Guide

**Mass Composition:** What's your element's % of mass after the reaction? Molar mass to obtain your value!

10. There are 3.75 moles of  $\text{CO}_2$ .

11. There are 3.75 moles of  $\text{CO}_2$ .

12. There are 3.75 moles of  $\text{CO}_2$ .

13. There are 3.75 moles of  $\text{CO}_2$ .

14. There are 3.75 moles of  $\text{CO}_2$ .

15. There are 3.75 moles of  $\text{CO}_2$ .

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23. There are 3.75 moles of  $\text{CO}_2$ .

24. There are 3.75 moles of  $\text{CO}_2$ .

25. There are 3.75 moles of  $\text{CO}_2$ .

26. There are 3.75 moles of  $\text{CO}_2$ .

Name \_\_\_\_\_ Period \_\_\_\_\_ Date \_\_\_\_\_

## The Mole Study Guide

Fill in the Blank boxes for the image below.



Write the Empirical Formula for each compound. Use subscripts to show the relative ratio of atoms.

| Molecular                           | Empirical |
|-------------------------------------|-----------|
| $\text{C}_6\text{H}_{12}\text{O}_6$ |           |
| $\text{C}_4\text{H}_8\text{O}_4$    |           |
| $\text{C}_2\text{H}_4\text{O}_2$    |           |
| $\text{C}_3\text{H}_6\text{O}_3$    |           |
| $\text{C}_5\text{H}_{10}\text{O}_5$ |           |

Calculate the Molar Mass for the following compounds.

10.  $\text{H}_2\text{O}$

11.  $\text{H}_2\text{O}_2$

12.  $\text{H}_2\text{O}$

13.  $\text{H}_2\text{O}_2$

Calculate the Percent Composition for each element in the following compound.

10.  $\text{H}_2\text{O}$

11.  $\text{H}_2\text{O}_2$

12.  $\text{H}_2\text{O}$

13.  $\text{H}_2\text{O}_2$

14.  $\text{H}_2\text{O}$

15.  $\text{H}_2\text{O}_2$

16.  $\text{H}_2\text{O}$

17.  $\text{H}_2\text{O}_2$

18.  $\text{H}_2\text{O}$

19.  $\text{H}_2\text{O}_2$

Determine the Empirical and Molecular Formula for the compound given the information below.

10. A compound has the following percent composition: 58.80% Hg, 21.87% Cl, 1.93% N. The molar mass for Aqueous is 520.91 g/mol.

a. If you have 100 g of Aqueous, how many grams of Hg, Cl, and N do you have?

b. Determine the Empirical Formula for Aqueous.

c. Determine the Molecular Formula for Aqueous.

# Study Guide The Mole

**Steven S. Zumdahl, Paul B. Kelter**



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