

Solubility Graph Reading

Name _____ Hr. _____

Use the solubility graphs to answer the following questions.

1) What mass (in grams) of KCl will dissolve in 100 grams of water at 70°C?

2) How many more or fewer grams of NaCl will dissolve in 100 grams of water at 70°C than grams of KCl in #1?

3) What mass of potassium nitrate will dissolve in 100 grams of water at 70°C?

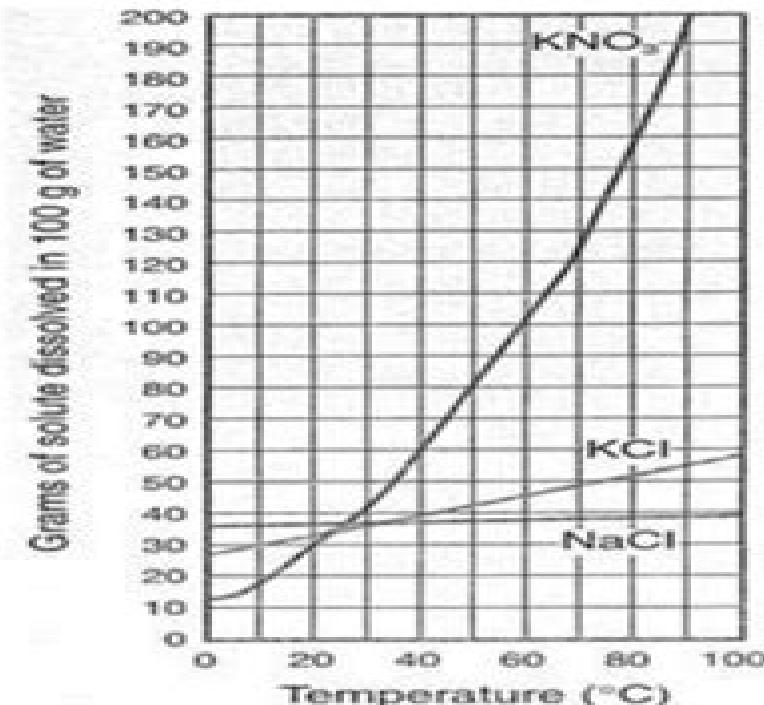
4) How much more potassium nitrate can be dissolved in 100 g of water that has been raised from 10°C to 50°C and originally contained the saturated amount?

5) 25g of potassium nitrate are dissolved in 100 grams of water and the temperature is raised to 90°C how much more chemical is required to form a saturated solution?

6) If the temperature of potassium chloride is dropped from 100°C to 40°C how much chemical will precipitate from solution if no supersaturated solution is formed?

7) Describe the relationship between temperature and solubility of a solid.

Effect of Temperature on the Solubility of Various Salts in Water



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Donald W. Shive

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