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Fall Boots Stagram Picks Ideas

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2 $\text{CO}_2 \text{ g} \rightarrow \text{O}_2 \text{ g}$ $\Delta h = 520 \text{ kJ}$
2 $\text{C}_2\text{H}_2 \text{ g} \rightarrow \text{C}_2\text{H}_6 \text{ g}$ $\Delta h = 945 \text{ kJ}$
 $\text{H}_2 \text{ g} \rightarrow \text{H}_2\text{O} \text{ g}$ $\Delta h = 285 \text{ kJ}$
 $\text{C}_2\text{H}_6 \text{ g} \rightarrow \text{C}_2\text{H}_2 \text{ g} + 3 \text{ H}_2 \text{ g}$ $\Delta h = 283 \text{ kJ}$
 $\text{O}_2 \text{ g} \rightarrow \text{O}_2 \text{ g}$ $\Delta h = 0 \text{ kJ}$

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3 $\text{O}_2 \text{ g} \rightarrow \text{O}_2 \text{ g}$ $\Delta h = 1411 \text{ kJ}$ mole
2 $\text{H}_2 \text{ g} \rightarrow \text{H}_2\text{O} \text{ g}$ $\Delta h = 1560 \text{ kJ}$ mole
 $\text{C}_2\text{H}_4 \text{ g} \rightarrow \text{C}_2\text{H}_6 \text{ g} + \text{H}_2 \text{ g}$ $\Delta h = 285 \text{ kJ}$ mole

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web thus for reaction 9 2 in which 1 mole of $h3o$ and 1 mole of oh combine to form 2 moles of $h2o$ h 55.8 kj as shown note that the enthalpy change h hproducts hreactants is positive if heat is absorbed that is if hproducts

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