

# Solubility Guide

TABLE 10.4 A solubility guide for common ionic compounds

Soluble ionic compounds	
Most compounds of the following ions are soluble...	Important exceptions...
Nitrate ( $\text{NO}_3^-$ )	None
Ethanoate ( $\text{CH}_3\text{COO}^-$ ) [common name: acetate]	None
Sodium, potassium ( $\text{Na}^+$ , $\text{K}^+$ )	None
Ammonium ( $\text{NH}_4^+$ )	None
Chloride ( $\text{Cl}^-$ )	$\text{AgCl}, \text{HgCl}_2$ , $\text{PbCl}_2$ is slightly soluble
Bromide ( $\text{Br}^-$ )	$\text{AgBr}, \text{HgBr}_2$ , $\text{PbBr}_2$
Iodide ( $\text{I}^-$ )	$\text{AgI}, \text{HgI}_2$ , $\text{PbI}_2$
Sulfate ( $\text{SO}_4^{2-}$ )	$\text{BaSO}_4$ , $\text{PbSO}_4$ , $\text{CaSO}_4$ is slightly soluble
Insoluble ionic compounds	
Most compounds of the following ions are insoluble...	Important exceptions...
Sulfide ( $\text{S}^{2-}$ )	$\text{Na}_2\text{S}, \text{K}_2\text{S}, \text{Li}_2\text{S}, (\text{NH}_4)_2\text{S}$
Carbonate ( $\text{CO}_3^{2-}$ )	$\text{Na}_2\text{CO}_3$ , $\text{K}_2\text{CO}_3$ , $\text{Li}_2\text{CO}_3$
Phosphate ( $\text{PO}_4^{3-}$ )	$\text{Na}_3\text{PO}_4$ , $\text{K}_3\text{PO}_4$ , $\text{Li}_3\text{PO}_4$
Hydroxide ( $\text{OH}^-$ )	$\text{NaOH}, \text{KOH}, \text{Ba}(\text{OH})_2$ , $\text{Ca}(\text{OH})_2$ is slightly soluble

# Solubility Guide

**Pradyot Patnaik**

## **Solubility Guide:**

Handbook of Aqueous Solubility Data Samuel H. Yalkowsky, Yan He, Parijat Jain, 2016-04-19 Over the years researchers have reported solubility data in the chemical pharmaceutical engineering and environmental literature for several thousand organic compounds Until the first publication of the Handbook of Aqueous Solubility Data this information had been scattered throughout numerous sources Now newly revised the second edition of CRC Handbook of Basic Tables for Chemical Analysis Thomas J. Bruno, Paris D.N. Svoronos, 2010-12-13 Winner of an Outstanding Academic Title Award for 2011 Researchers in organic chemistry chemical engineering pharmaceutical science forensics and environmental science make routine use of chemical analysis but the information these researchers need is often scattered in different sources and difficult to access The CRC Handbook of Basic Tables Solvent Recovery Handbook Ian McN. Smallwood, 2002 The pressure is on to cut plant emissions while still maintaining a cost effective operation Choosing the best solvent being aware of potential problems and the recovery of solvents has never been so important Traditionally solvents had been chosen on the basis of whether they can do the job effectively and economically However with regulations on exposure to solvent vapors becoming more stringent selecting the solvent that meets regulatory efficiency and economical criteria as early as possible in the process has become paramount Solvent Recovery Handbook Second Edition sets out the physical properties of the fifty most commonly used solvents The book supplies information on their behavior during and after use health and fire hazards the photochemical ozone creation potential POCP and recovery processes including practical aspects of the design and operation of batch stills It delivers state of the art coverage of every available recovery and disposal technology including removing solvents from gas water and residues separating used solvents and drying solvents What's more you'll find fact filled sections on the latest equipment safe effective operating procedures choosing solvents with recovery in mind and much more Updated and expanded Ian Smallwood's Solvent Recovery Handbook Second Edition hands you all the practical tools you need to efficiently and cost effectively process harmful organic solvents after re capture

**Laboratory Guide and Class Manual in Qualitative Chemical Analysis** Harry Sands Grindley, 1906 **Quality assurance of pharmaceuticals: a compendium of guidelines and related materials, tenth edition. Volume 1. Good practices and related regulatory guidance** World Health Organization, 2024-10-24 This publication represents a significant achievement in our ongoing effort to ensure that everyone can reach the highest possible level of health Over the last three decades we have seen the transformation of the pharmaceutical industry and the increasing intricacy of the product life cycle The challenges we face today are very different from those we faced when the first edition of this Compendium was published in 1997 However our mission remains the same to promote health keep the world safe and serve the vulnerable The new edition reflects the collective knowledge and expertise of countless professionals who have worked diligently to develop, revise and implement WHO guidelines for pharmaceuticals This includes experts from WHO Member States our Expert Advisory Panels

and Expert Committees on Specifications for Pharmaceutical Preparations and other organizations and has undergone extensive consultation with stakeholders worldwide This Compendium covers development through manufacturing and quality control to post marketing surveillance It provides a comprehensive framework for quality assurance that is both strong and flexible capable of meeting the requirements of a rapidly changing global health landscape The 10th edition is a collection of knowledge and tools for empowerment enabling all stakeholders in the pharmaceutical industry to make informed decisions that prioritize patient safety and well being The Chemistry of Synthetic Dyes V8 K

Venkataraman,2012-12-02 The Chemistry of Synthetic Dyes Volume VII stresses the relation between the chemistry of synthetic dyes and their application properties This book describes the dyes for leather synthetic carotenoids as food colorants and solvent dyes The phenomenal progress made in transfer printing which the first commercial process became available as recently as 1968 is also discussed This text likewise considers the influence of structural factors on the lightfastness of dyed fibers and structures of dyes with their technical properties This volume is a good reference for organic chemists and technologists working on the synthesis of dyes and their applications **A Guide to Materials**

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**Solubilities of Inorganic and Organic Substances** ,1911     *Drug Metabolism Handbook* Ala F. Nassar,2009-01-28 A valuable reference tool for professionals involved in the industry Drug Metabolism in Pharmaceuticals covers new tools such as LC MS and LC MS NMR along with experimental aspects of drug metabolism This work fills a gap in the literature by

covering the concepts and applications of pharmaceutical research development and assessment from the point of view of drug metabolism By providing both a solid conceptual understanding of the drug metabolism system and a well illustrated detailed demonstration and explanation of cutting edge tools and techniques this book serves as a valuable reference tool for bench scientists medical students and students of general health sciences

**Study Guide [to] Chemistry, a Systematic Approach [by] Sisler, Dresdner, and Mooney** William H. Myers, William T. Mooney, 1980

Handbook of Solubility Data for Pharmaceuticals Abolghasem Jouyban, 2009-08-26 Aqueous solubility is one of the major challenges in the early stages of drug discovery One of the most common and effective methods for enhancing solubility is the addition of an organic solvent to the aqueous solution Along with an introduction to cosolvency models the Handbook of Solubility Data for Pharmaceuticals provides an extensive datab

*Laboratory Guide to Pharmaceutical Technique* University of Wisconsin. Department of Pharmacy, 1926 **Report** Great Britain. Ministry of Agriculture and Fisheries. Research and Education Division, 1928 **A**

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