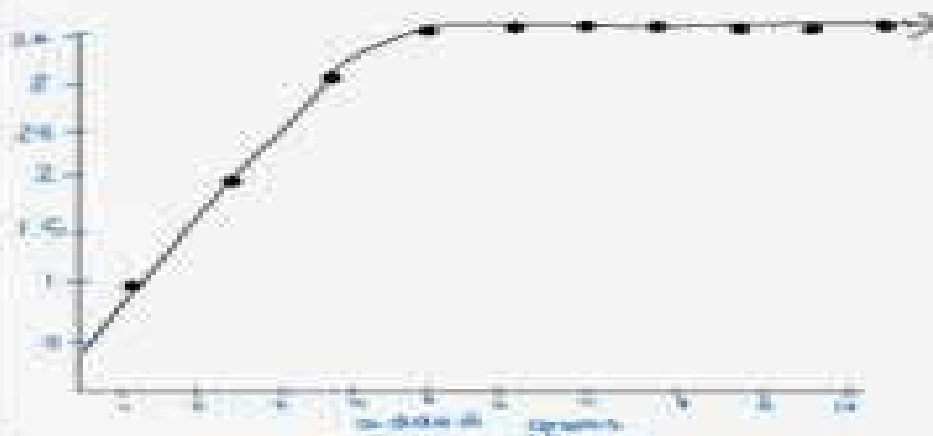


Model 2 – Solute Dissolved vs. Solute Added

The following data refer to an experiment in which a measured mass of solid is added to 10.0 g of 20 °C water. The mixture is stirred and allowed to sit for 5 hours. Ten separate trials are conducted for the experiment.

Trial Number	Mass of solute added (grams)	Mass of solute dissolved (grams)
1	1.0	1.0
2	2.0	2.0
3	3.0	3.0
4	4.0	3.6
5	5.0	3.6
6	6.0	3.6
7	7.0	3.6
8	8.0	3.6
9	9.0	3.6
10	10.0	3.6



11. Four of the trials in Model 2 correspond to beakers A, B, D, and E from Model 1. Write the letters for these beakers next to the corresponding trial numbers in Model 2.

12. Identify the following variables in the experiment in Model 2.

Dependent variable

mass of solute
that was dissolved

Independent variable

gms of solute
added

Controlled variable(s)

temp. of water, mass of water,
stirring

13. Sketch a graph of the data for the experiment in Model 2. A space has been provided next to the data table. Be sure to consider which variable belongs on each axis.

14. Consider the data in Model 2.

- a. Which trials represent solutions that are unsaturated?

1-3-4-6-8-9

- b. Which trial numbers represent solutions that are saturated?

4-5-6-7-8-9-10

- c. Describe the feature in the graph that can help you identify the saturated solutions. Explain.

saturation solutions are at a constant level

Solubility Pogil Answer Key

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The Answer Key Rachel Turoscy, 2018-08-09 The Answer Key A Comprehensive Explanation of Problem Solving Methods for General Chemistry Success Volume 1 is a concise and accessible textbook that covers the critical information a student needs to understand the basic mathematics used in chemistry courses The book provides easy to understand step by step instructions for solving general chemistry problems The book begins with chapters dedicated to problem solving methodology and unit conversions In subseq **A Calorimetric Test of the Solubility Equation** Robert D. Vold, 1935

Standard Test Method for Measurements of Aqueous Solubility, *Standard Test Method for Solubility of Fixed Gases in Liquids*, **Solutions** American Association for the Advancement of Science, 1975 Construct a hypothesis to explain observations made or data collected concerning the effects of temperature and the nature of the solvent on the solubility of solids to demonstrate methods to manipulate and hold constant variables in a test of a hypothesis and to identify data that support and do not support a hypothesis *Standard Test Method for Solubility of Fixed Gases in Low-Boiling Liquids*, **Solubility Data Series**, 1996 **Standard Test Methods for Estimation of Solubility of Gases in Petroleum and Other Organic Liquids** ASTM International, 2012 Solubility Data for U(VI) Hydroxide and Np(VI) Hydrous Oxide: Application of MCC-3 Methodology D. Rai K.M. Krupka (R.W. Fulton, R.G. Strickert), 1984 *Standard Test Method for Solubility of Fixed Gases in Low-Boiling Liquids Withdrawn 2017*,

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