

Solubility Curve Worksheet

- 1) Define solubility.
- 2) Look at the graph below. In general, how does temperature affect solubility?

- 3) Which compound is LEAST soluble at 10 °C? _____
- 4) How many grams of KCl can be dissolved in 100g of water at 80°C? _____
- 5) How many grams of NaCl can be dissolved in 100g of water at 90°C? _____
- 6) At 40°C, how much KNO₃ can be dissolved in 100g of water? _____
- 7) Which compound shows the least amount of change in solubility from 0°C-100°C?

- 8) At 30°C, 90g of NaNO₃ is dissolved in 100g of water. Is this solution saturated or unsaturated?

- 9) At 60°C, 72g of NH₄Cl is dissolved in 100g of water. Is this solution saturated or unsaturated?

- 10) A saturated solution of KClO₃ is formed from one hundred grams of water. If the saturated solution is cooled from 90°C to 50°C, how many grams of precipitate are formed?

- 11) A saturated solution of NH₄Cl is formed from one hundred grams of water. If the saturated solution is cooled from 80°C to 40°C, how many grams of precipitate are formed?

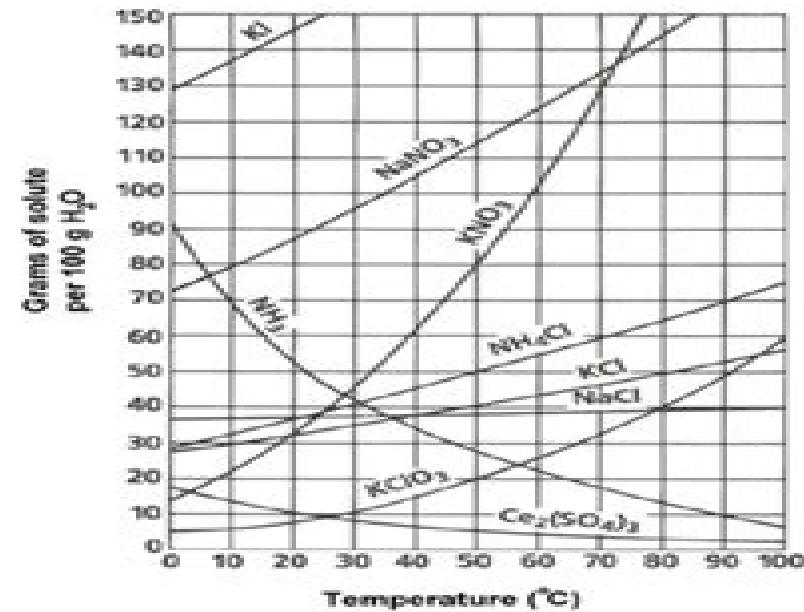
- 12) Which compounds show a decrease in solubility from 0°C-100°C?

- 13) Which compound is the most soluble at 10°C?

- 14) Which compound (besides Ce₂(SO₄)₃) is the least soluble at 50°C?

- 15) For each of the following solutions, explain how much of the solute will dissolve and how much will remain undissolved at the bottom of the test tube?
 - a) 120 g of KCl in 100 g of water at 80°C

 - b) 130 g of NaNO₃ in 100 g of water at 50°C



Solubility Curve And Lab Answers

Flavius Josephus

Solubility Curve And Lab Answers:

Practical Chemistry Labs Leonard Saland,1989 Features self contained step by step activities using common materials and covering topics from food chemistry to papermaking and electrochemistry Illustrates the connection between the real world and chemistry concepts such as solutions chemistry acids and bases and more Includes teacher notes quizzes and answers to help monitor student progress

Science Educator's Guide to Laboratory Assessment Rodney L. Doran,2002 The book opens with an up to date discussion of assessment theory research and uses Then comes a wealth of sample assessment activities in biology chemistry physics and Earth science Keyed to the National Science Education Standards the activities include reproducible task sheets and scoring rubrics All are ideal for helping students reflect on their own learning during science lab

A Laboratory Outline of General Chemistry Herbert Newby McCoy,Ethel Mary Terry,1920 Chemistry in the Community American Chemical Society,2006-01-31 This laboratory based text centres itself around decision making activities where students apply their chemistry knowledge to realistic situations This fifth edition includes more photographs new drawings and new design

A Laboratory Outline of General Chemistry Alexander Smith,1907 *Laboratory exercises in general chemistry* William Foster,1925 *A Laboratory Outline of College*

Chemistry Alexander Smith,1916 **Practical Chemistry for CSEC** Norman Lambert,1987-03-30 Practical Chemistry is a unique practice book for CXC It provides a wealth of revision exercises and a guide to all the detailed experimental work covered in the CXC Chemistry syllabus Section A Practical guidance for teachers and classes perform

A Laboratory Outline of Smith's College Chemistry James Kendall,1929 *The Laboratory Study of Chemistry* Herbert Royl Smith,1918

Contemporary Chemistry Leonard Saland,1993 This comprehensive guide gives you lesson plans activities and tests for two sequential semester long chemistry courses It is designed to work with our student book Contemporary Chemistry Each lesson plan features a DO NOW section to engage students as soon as they get to class instructional objectives an aimfor that class period a motivational application questions or demonstrations to help students draw valid conclusions homework assignments You also get term calendars weekly tests and complete answer keys

College Chemistry Companion James Kendall,1924 *Selected Readings in General Chemistry* William Franklin Kieffer,Robert K. Fitzgerel,1959 Experiments in General Chemistry Hosmer Ward Stone,Max Shaw Dunn,1929

14th International Symposium on Industrial Crystallization ,1999 This conference provides a forum for discussion of the advances in the theory and practice of crystallization as it relates to the production of bulk crystalline materials *The genuine works of Flavius Josephus: Containing four books of the Jewish war* Flavius Josephus,1920 *Chemistry* Lemay,Prentice-Hall Staff,1996 *Scientific American* ,1902 **Fundamentals of Chemistry Laboratory Studies** Frank Brescia,1980 **Review of the Electrical Communication Laboratory** Denki Tsūshin Kenkyūjo (Japan),1963 Includes abstracts of the laboratory's Technical journal and other articles

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