



Solubility Study Guide

Name _____

Teacher: _____

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References

Chemistry in the Community pp 45-65
Water properties concept map
Solubility concept map
Solubility, graphing and analysis
Solubility Practice: A Curvy Subject
Study Solvent Lab

Vocabulary

Understand and apply the definitions or symbols of the following terms.

Colligative properties
Distillation
Filtration
Homogeneous
Mixtures
Nonpolar
Polar
Pure substances
Solubility

Solute
Solvent
Solution
Supersaturated
Saturated
Unsaturated
 δ^+ and δ^-

Targeted Skills

1. Describe the structure of water and relate its unusual physical properties to its structure, high boiling point, surface tension, high specific heat.
2. Draw the structure of water and relate its structure to the ability of water to act as a universal solvent.
3. Compare/contrast the solubilities of various ionic and covalent solutes.
4. Apply: Like dissolves like:
Polar solvents dissolve Polar and Ionic solutes
Nonpolar solvents dissolve Nonpolar solutes
5. Describe factors that affect the rate of dissolving, such as particle size, stirring, temperature, nature of the solid solutes.
6. Interpret information from a solubility curve (graph).

Make sure you remember from past tests?

Atomic structure
Significant figures
Accuracy & precision

Equations:(extension)

Solution concentration = $\frac{\text{amount of solute}}{\text{amount solute} + \text{solvent}}$

Test Breakdown

30 Multiple-Choice Problems (max.) – 60%

- 1 Free Response Problem chosen from the Free Response Test Bank – 10%
- 1 Open-Ended Problem chosen from the Open-Ended Test Bank – 10%
- 1 Performance Assessment – 20%

Solubility Study Guide

Karen C. Timberlake



Solubility Study Guide:

Study Guide for Chemical Principles [by] Steven S. Zumdahl Steven S. Zumdahl, Paul B. Kelter, 1995 The Study Guide reflects the unique problem solving approach taken by the Chemical Principles text The new edition of the Study Guide includes many new worked out examples *Study Guide to Chemistry and Life* John William Hill, Dorothy M. Feigl, 1978

Hazmat Chemistry Study Guide (Second Edition) Jill Meryl Levy, 2011 **Study Guide and Solutions Manual, Fundamentals of General, Organic, and Biological Chemistry, Third Edition** John McMurry, Susan McMurry, 1999 Provides worked out solutions to text problems along with chapter by chapter outlines and a variety of self tests at the end of each chapter **The Practice of Chemistry Study Guide & Solutions Manual** Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises **General Chemistry, Study Guide** James E. Brady, 1990-03-01

The Fifth Edition retains the pedagogical strengths that made the previous editions so popular and has been updated reorganized and streamlined Changes include more accessible introductory chapters with greater stress on the logic of the periodic table earlier introduction of redox reactions greater emphasis on the concept of energy a new section on Lewis structures earlier introduction of the ideal gas law and a new development of thermodynamics Each chapter ends with review questions and problems **Study Guide for Chang's Chemistry** Kenneth W. Watkins, 1988 *Robinson Chemistry Study Guide* Robinson, 1992 **Study Guide and Partial Solutions Manual, Fundamentals of General, Organic, and Biological Chemistry** Susan McMurry, 1996 This internationally acclaimed detective series is just the thing for lovers of those Number One Ladies looking for a darker more realistic view of Botswana Sue Baker Publishing News *General, Organic, and Biological Chemistry Study Guide and Selected Solutions* Karen C. Timberlake, 2001-11 Keyed to the learning goals in the text this guide is designed to promote active learning through a variety of exercises with answers and mastery exams The guide also contains complete solutions to odd numbered problems **Study Guide and Glossary to Accompany Chemical Principles** Daniel L. Reger, Edward E. Mercer, Robert S. Boikess, 1978 **Student Study Guide and Solutions Manual to Accompany General, Organic, and Biochemistry** Katherine J. Denniston, Denniston Katherine, 2006 *Chemistry, Study Guide* Bernice G. Segal, Peter S. Shenkin, 1989-02-14 This Second Edition of the first year chemistry text known for its clarity of exposition and its large number of illustrative worked problems contains a more rigorous treatment of electrochemistry chemical equilibrium and thermochemistry Worked examples now number over 300 and exercises over 1460 *Student Study Guide and Solutions Manual to accompany General Organic and Biological Chemistry, 1e* Kenneth W. Raymond, 2005-10-07 Finally readers have a shorter less intimidating introduction to general organic and biological chemistry Not only is Raymond's text concise it also takes an integrated approach to presenting important topics in a way that makes the material easier to understand In this approach similarities can be exploited and

concepts reinforced The result is that readers see the strong connections that exist between these three branches of chemistry *Organic Chemistry, 5e Student Study Guide and Solutions Manual* David R. Klein, Laurie S. Starkey, 2025-03-04

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