

Solubility Rules

When salt solutions are mixed, two things can happen: 1) no reaction (all ions remain soluble), or 2) a solid forms (called a precipitate). You will examine which salts (positive + negative ions) are soluble and which are insoluble.

Safety: Wear goggles, wash your hands after lab.

Hypothesis: Look at the grid below. Looking back at your solubility rules sheet, which of these reactions do you expect to form a precipitate (solid)? _____

Procedure

1. Obtain from teacher: 1 well plate and aqueous solutions to be used in investigations.
2. Place your well plate on the lab bench.
3. Using the dropper bottles at your lab station, carefully add one drop (from a height of 1-2 cm) to each well for that ion. E.g. for Br^- a drop goes in each box under Br^- ; for Ag^+ a drop goes in each box left of Ag^+ . (be careful to note the label on each bottle). It is critical to add drops from a height of 1-2 cm. If you touch the tip of the dropper bottle to another solution, the contents of that bottle can become contaminated, ruining the results of your group and other students. Continue this procedure until all ions have been added together. At the end of this step you should have, in each box, a mixture of two drops (1 positive ion + 1 negative ion).
4. Look in the boxes: some will contain a precipitate (i.e. solid), others will be clear (indicating the salt is soluble).
5. On the grid below, place an "X" in every box where a precipitate has formed.
6. Rinse off the reaction grid in the sink. Rinse out the sink. Wipe the grid clean using a damp paper towel. Dry the grid. Return all equipment. Wipe off your lab bench. Wash your hands thoroughly.

	AgNO_3 (Ag^+)	$\text{Pb}(\text{NO}_3)_2$ (Pb^{+2})	CaCl_2 (Ca^{+2})	CuSO_4 (Cu^{+2})	$\text{Fe}(\text{NO}_3)_3$ (Fe^{+3})	K_2CrO_4 (K^+)
Na_3PO_4 (PO_4^{-3})	a	d	g	j	m	p
NaOH (OH^-)	b	e	h	k	n	q
Na_2SO_4 (SO_4^{-2})	c	f	i	l	o	r

Solubility Rules Lab Answer Key

Jane Butler Kahle

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

Solubility Rules Lab Answer Key:

General, Organic, and Biological Chemistry Study Guide and Selected Solutions Karen C. Timberlake, 2001-11
Keyed to the learning goals in the text this guide is designed to promote active learning through a variety of exercises with answers and mastery exams The guide also contains complete solutions to odd numbered problems **You Want Me to Teach What?** Norman Joseph LaFave, 2012 Problem You feel shaky about being assigned to teach upper level science and math and need to get up to speed fast Solution Follow this concise book s tried and true methods which you can integrate into your classroom and lesson plans starting from the first day of class You Want Me to Teach What avoids long discussions of education theory and specific lesson plans Instead it concentrates on general techniques for approaching a variety of problems and enhancing your teaching skills in science and math It covers student psychology classroom management planning instruction problem solving techniques laboratory methods and reporting assessment and professional development Without feeling inundated you ll find a wealth of sensible guidance whether you re a preservice education major wanting to teach physical science or mathematics a new teacher looking for practical methods to integrate into your instruction or an experienced teacher in search of fresh ways to improve in the classroom **Cooperative Chemistry Lab Manual** Cooper, 2005-02 The laboratory course described in the lab manual emphasizes experimental design data analysis and problem solving Inherent in the design is the emphasis on communication skills both written and oral Students work in groups on open ended projects in which they are given an initial scenario and then asked to investigate a problem There are no formalized instructions and students must plan and carry out their own investigations **Teaching High School Science Through Inquiry and Argumentation** Douglas Llewellyn, 2013 For Grades 9 12 this new edition covers assessment questioning techniques to promote learning new approaches to traditional labs and activities that emphasize making claims and citing evidence **Teaching High School Science Through Inquiry** Douglas Llewellyn, 2005 This is the secondary school l version of Llewellyn s strong Corwin debut *Inquire Within Implementing Inquiry Based Science Standards* 2000 This book focuses on raising a teacher s capacity to teach science through an inquiry based process implementing inquiry as stated by the national standards Microscale and Miniscale Organic Chemistry Laboratory Experiments Allen M. Schoffstall, Barbara A. Gaddis, Melvin L. Druehinger, 2000 This work offers a comprehensive introductory treatment of the organic laboratory techniques for handling glassware and equipment safety in the laboratory micro and mini scale experimental procedures theory of reactions and techniques applications and spectroscopy Resources in Education , 1987 *Hydrology and Water Resources of Small Islands* Emilio Custodio, International Hydrological Programme. IHP-III. Project 4.6, 1991 *Yoga Journal* , 1982-05 For more than 30 years Yoga Journal has been helping readers achieve the balance and well being they seek in their everyday lives With every issue Yoga Journal strives to inform and empower readers to make lifestyle choices that are healthy for their bodies and minds We are dedicated to

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Password Card (Chapters 1 And 8 - 13) Bill W. Tillery,Tillery Bill,2004-01-09

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