

## EJERCICIO PROPUESTO 7.5.

100 Kg de una disolución acuosa de acetona de composición 35% en peso de acetona se tratan con metil-isobutil-cetona a 30°C en un contacto sencillo de extracción. Calcular:

a) Cantidades mínima y máxima de disolvente a emplear.

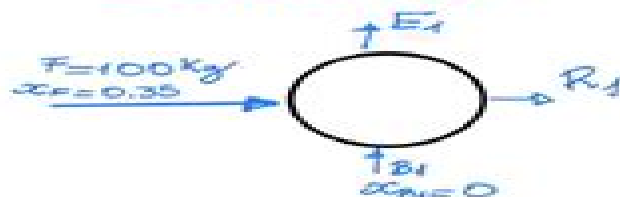
R:  $B_{min}=6.38 \text{ Kg}$ ,  $B_{max}=2817 \text{ Kg}$

b) Concentraciones de los productos extraído y refinado si se emplean 80 Kg de metil-isobutil-cetona.

R:  $Y_I=0.843$ ,  $X_I=0.142$

c) Porcentaje de acetona separada en el producto extraído.

R: %Acetona=71.4%



Empezamos con la cantidad mínima

Balance de masa total

$$F + B_{min} = D$$

Balance de masa en soluto

$$F x_F + B_{min} x_B = D x_D$$

$$B_{min} = \frac{F(x_F - x_D)}{x_D}$$

$$B_{min} = \frac{(100)(0.35 - 0.329)}{0.329}$$

$$B_{min} = 6.38 \text{ Kg}$$



Para la cantidad máxima ubicamos el punto G

$$\text{Balance total} \rightarrow F + B_{max} = G$$

$$\text{Balance en soluto} \rightarrow F x_F + B_{max} x_B = G x_G$$

$$B_{max} = \frac{F(x_F - x_G)}{x_G}$$

$$B_{max} = \frac{(100)(0.35 - 0.012)}{0.012}$$

$$B_{max} = 2816.7 \text{ Kg}$$



# Solucionario Ocon Tojo

**CH Cherryholmes**



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