

EJERCICIO PROPUESTO 7.5.

100 Kg de una disolución acuosa de acetona de composición 35% en peso de acetona se tratan con metil-isobutil-cetona a 30°C en un contacto sencillo de extracción. Calcular:

a) Cantidades mínima y máxima de disolvente a emplear.

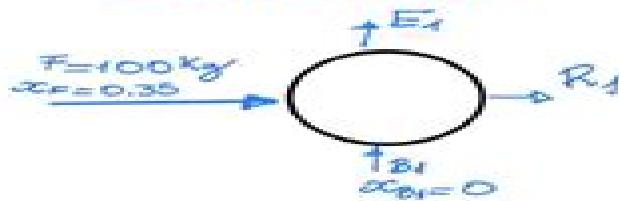
R: $B_{min}=6.38 \text{ Kg}$, $B_{max}=2817 \text{ Kg}$

b) Concentraciones de los productos extraído y refinado si se emplean 80 Kg de metil-isobutil-cetona.

R: $y_1=0.843$, $x_1=0.142$

c) Porcentaje de acetona separada en el producto extraído.

R: $\% \text{Acetona}=71.4\%$



Empiezamos con la cantidad mínima

Balance de masa total

$$F + B_{min} = D$$

Balance de masa en soluto

~~$F_{sol} + B_{min} = D_{sol}$~~

$$B_{min} = F(x_F - x_{sol})$$

$$B_{min} = \frac{100}{0.843} (0.35 - 0.329)$$

B_{min} = 6.38 Kg



Para la cantidad máxima ubicamos el punto G

Balance total $\rightarrow F + B_{max} = G$

Balance en soluto $\rightarrow F_{sol} + B_{max} = G_{sol}$

~~$B_{max} = \frac{F(x_F - x_{sol})}{x_{sol}}$~~

$$B_{max} = \frac{100}{0.012} (0.35 - 0.012)$$

B_{max} = 2816.7 Kg



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Table of Contents Solucionario Ocon Tojo

1. Understanding the eBook Solucionario Ocon Tojo
 - The Rise of Digital Reading Solucionario Ocon Tojo
 - Advantages of eBooks Over Traditional Books
2. Identifying Solucionario Ocon Tojo
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Solucionario Ocon Tojo
 - User-Friendly Interface
4. Exploring eBook Recommendations from Solucionario Ocon Tojo
 - Personalized Recommendations
 - Solucionario Ocon Tojo User Reviews and Ratings
 - Solucionario Ocon Tojo and Bestseller Lists
5. Accessing Solucionario Ocon Tojo Free and Paid eBooks

- Solucionario Ocon Tojo Public Domain eBooks
- Solucionario Ocon Tojo eBook Subscription Services
- Solucionario Ocon Tojo Budget-Friendly Options

6. Navigating Solucionario Ocon Tojo eBook Formats

- ePUB, PDF, MOBI, and More
- Solucionario Ocon Tojo Compatibility with Devices
- Solucionario Ocon Tojo Enhanced eBook Features

7. Enhancing Your Reading Experience

- Adjustable Fonts and Text Sizes of Solucionario Ocon Tojo
- Highlighting and Note-Taking Solucionario Ocon Tojo
- Interactive Elements Solucionario Ocon Tojo

8. Staying Engaged with Solucionario Ocon Tojo

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Solucionario Ocon Tojo

9. Balancing eBooks and Physical Books Solucionario Ocon Tojo

- Benefits of a Digital Library
- Creating a Diverse Reading Collection Solucionario Ocon Tojo

10. Overcoming Reading Challenges

- Dealing with Digital Eye Strain
- Minimizing Distractions
- Managing Screen Time

11. Cultivating a Reading Routine Solucionario Ocon Tojo

- Setting Reading Goals Solucionario Ocon Tojo
- Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Solucionario Ocon Tojo

- Fact-Checking eBook Content of Solucionario Ocon Tojo
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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