

# Solution Concentration

## Study Guide

### Key Concepts

- Concentrations can be measured qualitatively and quantitatively.
- Molarity is the number of moles of solute dissolved per liter of solution.

$$\text{molarity (M)} = \frac{\text{moles of solute}}{\text{liters of solution}}$$

- Molality is the ratio of the number of moles of solute dissolved in 1 kg of solvent.

$$\text{molality (m)} = \frac{\text{moles of solute}}{\text{kg of solvent}}$$



# Solution Concentration Study Guide Key

**J.C. Cummins, Jack D. Cummins**

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*Student Study Guide and Solutions Manual to Accompany General, Organic, and Biochemistry* Katherine J. Denniston, Denniston Katherine, 2006    **Study guide to accompany chemistry** Karen Timberlakes, 1983    *Study Guide/Selected Solutions Manual* Julie R. Frentrup, 2002-06 Contains a brief overview of every chapter review of skills self tests and the answers and detailed solutions to all odd numbered end of chapter problems in the text book    *Students Study Guide* Fredeen, Pearson, 2003-07    **Study Guide with Selected Solutions** Karen Timberlake, 2002-08-06 Written by the author the Study Guide is keyed to the learning goals in the text and designed to promote active learning through a variety of exercises with answers and mastery exams Also contains complete solutions to odd numbered problems

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