

# Acids & Bases Study Guide

NAME \_\_\_\_\_

DATE \_\_\_\_\_

PERIOD \_\_\_\_\_

1. Write the chemical formula for each acid.

2. Write the chemical formula for each base.

3. Write the chemical formula for each salt.

4. Write the chemical formula for each acid.

5. Write the chemical formula for each base.

6. Write the chemical formula for each salt.

7. Write the chemical formula for each acid.

8. Write the chemical formula for each base.

9. Write the chemical formula for each salt.

10. Write the chemical formula for each acid.

11. Write the chemical formula for each base.

12. Write the chemical formula for each salt.

NAME \_\_\_\_\_ DATE \_\_\_\_\_ PERIOD \_\_\_\_\_

**Acid-Base Reactions**

Acid	Base	Salt	Water
$\text{HCl}$	$\text{NaOH}$	$\text{NaCl}$	$\text{H}_2\text{O}$
$\text{H}_2\text{SO}_4$	$\text{KOH}$	$\text{K}_2\text{SO}_4$	$\text{H}_2\text{O}$
$\text{HNO}_3$	$\text{Ca(OH)}_2$	$\text{Ca(NO}_3)_2$	$\text{H}_2\text{O}$
$\text{H}_3\text{PO}_4$	$\text{Mg(OH)}_2$	$\text{Mg}_3(\text{PO}_4)_2$	$\text{H}_2\text{O}$
$\text{H}_2\text{CO}_3$	$\text{Na}_2\text{CO}_3$	$\text{Na}_2\text{HCO}_3$	$\text{H}_2\text{O}$
$\text{H}_2\text{S}$	$\text{Fe(OH)}_3$	$\text{Fe}_2(\text{S}_2)_3$	$\text{H}_2\text{O}$
$\text{H}_2\text{SO}_3$	$\text{Al(OH)}_3$	$\text{Al}_2(\text{SO}_3)_3$	$\text{H}_2\text{O}$
$\text{H}_2\text{SiO}_3$	$\text{Ba(OH)}_2$	$\text{BaSiO}_3$	$\text{H}_2\text{O}$
$\text{H}_2\text{B}_4\text{O}_7$	$\text{LiOH}$	$\text{Li}_2\text{B}_4\text{O}_7$	$\text{H}_2\text{O}$
$\text{H}_2\text{C}_2\text{O}_4$	$\text{NH}_4\text{OH}$	$\text{NH}_4\text{C}_2\text{O}_4$	$\text{H}_2\text{O}$
$\text{H}_2\text{C}_2\text{O}_3$	$\text{H}_2\text{O}$	$\text{H}_2\text{C}_2\text{O}_4$	$\text{H}_2\text{O}$

1. Write the chemical formula for each acid.

2. Write the chemical formula for each base.

3. Write the chemical formula for each salt.

4. Write the chemical formula for each acid.

5. Write the chemical formula for each base.

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10. Write the chemical formula for each acid.

11. Write the chemical formula for each base.

12. Write the chemical formula for each salt.

# Study Guide Numeration Bases

**Educational Testing Service**

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## **Table of Contents Study Guide Numeration Bases**

1. Understanding the eBook Study Guide Numeration Bases
  - The Rise of Digital Reading Study Guide Numeration Bases
  - Advantages of eBooks Over Traditional Books
2. Identifying Study Guide Numeration Bases
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Study Guide Numeration Bases
  - User-Friendly Interface
4. Exploring eBook Recommendations from Study Guide Numeration Bases
  - Personalized Recommendations
  - Study Guide Numeration Bases User Reviews and Ratings
  - Study Guide Numeration Bases and Bestseller Lists
5. Accessing Study Guide Numeration Bases Free and Paid eBooks
  - Study Guide Numeration Bases Public Domain eBooks
  - Study Guide Numeration Bases eBook Subscription Services
  - Study Guide Numeration Bases Budget-Friendly Options

6. Navigating Study Guide Numeration Bases eBook Formats
  - ePub, PDF, MOBI, and More
  - Study Guide Numeration Bases Compatibility with Devices
  - Study Guide Numeration Bases Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Study Guide Numeration Bases
  - Highlighting and Note-Taking Study Guide Numeration Bases
  - Interactive Elements Study Guide Numeration Bases
8. Staying Engaged with Study Guide Numeration Bases
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Study Guide Numeration Bases
9. Balancing eBooks and Physical Books Study Guide Numeration Bases
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Study Guide Numeration Bases
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Study Guide Numeration Bases
  - Setting Reading Goals Study Guide Numeration Bases
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Study Guide Numeration Bases
  - Fact-Checking eBook Content of Study Guide Numeration Bases
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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