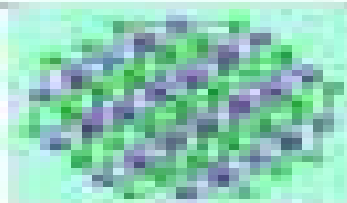


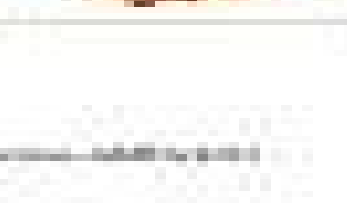


## Unit 8 – Part 2: Lesson 3

### Unit 8: Science Subtopic 3

In this table below identify the type of solid formed, whether ionic, molecular, metallic, network, etc. Then predict its colour, phase (solid, liquid or gas) and whether it would conduct in the liquid or in the solid state. Please

| Chemical formula | Crystal structure   | Type          | Boiling point (°C) | Physical state at room temperature   |
|------------------|---|---------------|--------------------|--|
| $\text{NaCl}$    |    | ionic         | 1413               | solid<br>white<br>brittle<br>high melting point<br>conducts in liquid<br>does not conduct in solid |
| $\text{CO}_2$    |    | molecular     | -78.5              | solid<br>white<br>brittle<br>low melting point<br>does not conduct in liquid or solid              |
| $\text{Fe}$      |   | metallic      | 2750               | solid<br>shiny<br>ductile<br>conducts in liquid and solid  |
| $\text{Si}$      |  | network solid | 2355               | solid<br>grey<br>brittle<br>conducts in liquid and solid   |

## Unit 6 Worksheet 2 Why Structure Is Important Answers

**Ritsuko Nakata, Karen Frazier, Barbara  
Hoskins, Carolyn Graham**



## Unit 6 Worksheet 2 Why Structure Is Important Answers:

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