

General, Organic & Biological Chemistry, 5e (Timberlake)
Chapter 7 Chemical Reactions and Quantities

7.1 Multiple-Choice Questions

- 1) A chemical equation is balanced when
A) the total number of molecules is the same in reactants and products.
B) the total number of ions is the same in reactants and products.
C) the sum of the coefficients of the reactants is equal to the sum of the coefficients of the products.
D) the number of atoms of each element is the same in reactants and products.
E) the charge on each atom is the same in reactants and products.

Answer: D

- 2) In any balanced chemical equation, the number of each type of atom on both sides of the equation is

- A) doubled.
B) the same.
C) decreased by one.
D) increased by one.
E) dependent on the temperature.

Answer: B

- 3) Which of the following gives the balanced equation for this reaction?



- A) $\text{KPO}_4 + \text{CaNO}_3 + \text{KNO}_3$
B) $\text{K}_3\text{PO}_4 + \text{Ca}(\text{NO}_3)_2 \rightarrow \text{Ca}_3(\text{PO}_4)_2 + 3\text{KNO}_3$
C) $2\text{K}_3\text{PO}_4 + \text{Ca}(\text{NO}_3)_2 \rightarrow \text{Ca}_3(\text{PO}_4)_2 + 6\text{KNO}_3$
D) $2\text{K}_3\text{PO}_4 + 3\text{Ca}(\text{NO}_3)_2 \rightarrow \text{Ca}_3(\text{PO}_4)_2 + 6\text{KNO}_3$
E) $\text{K}_3\text{PO}_4 + \text{Ca}(\text{NO}_3)_2 \rightarrow \text{Ca}_3(\text{PO}_4)_2 + \text{KNO}_3$

Answer: D

- 4) Which of the following correctly gives the best coefficients for the reaction below?



- A) 1, 1, 1, 1
B) 1, 2, 1, 4
C) 2, 4, 2, 8
D) 1, 4, 1, 4
E) 2, 4, 2, 4

Answer: B

- 5) What coefficient is placed in front of O_2 to complete the balancing of the following equation?



- A) 1
B) 3
C) 5
D) 7
E) 9

Answer: D

- 6) What is the coefficient of hydrogen, H_2 , when the following equation is balanced?



- A) 1
B) 2
C) 3
D) 4
E) 5

Answer: C

- 7) In this reaction, what is the coefficient for calcium oxide? $\text{CaO}(s) + \text{CO}_2(g) \rightarrow$

Timberlake Chemistry Practice Test Ch 07

Karen Timberlake



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