

General, Organic & Biological Chemistry, 5e (Timberlake)
Chapter 7 Chemical Reactions and Quantities

7.1 Multiple-Choice Questions

1) A chemical equation is balanced when

- A) the total number of molecules is the same in reactants and products.
- B) the total number of ions is the same in reactants and products.
- C) the sum of the coefficients of the reactants is equal to the sum of the coefficients of the products.
- D) the number of atoms of each element is the same in reactants and products.
- E) the charge on each atom is the same in reactants and products.

Answer: D

2) In any balanced chemical equation, the number of each type of atom on both sides of the equation is

- A) doubled.
- B) the same.
- C) decreased by one.
- D) increased by one.
- E) dependent on the temperature.

Answer: B

3) Which of the following gives the balanced equation for this reaction?

$$K_3PO_4 + Ca(NO_3)_2 \rightarrow Ca_3(PO_4)_2 + KNO_3$$

- A) $KPO_4 + CaNO_3 + KNO_3$
- B) $K_3PO_4 + Ca(NO_3)_2 \rightarrow Ca_3(PO_4)_2 + 3KNO_3$
- C) $2K_3PO_4 + Ca(NO_3)_2 \rightarrow Ca_3(PO_4)_2 + 6KNO_3$
- D) $2K_3PO_4 + 3Ca(NO_3)_2 \rightarrow Ca_3(PO_4)_2 + 6KNO_3$
- E) $K_3PO_4 + Ca(NO_3)_2 \rightarrow Ca_3(PO_4)_2 + KNO_3$

Answer: D

4) Which of the following correctly gives the best coefficients for the reaction below?



- A) 1, 1, 1, 1
- B) 1, 2, 1, 4
- C) 2, 4, 2, 8
- D) 1, 4, 1, 4

Answer: B

5) What coefficient is placed in front of O_2 to complete the balancing of the following equation?



- A) 1
- B) 3
- C) 5
- D) 7
- E) 0

Answer: D

6) What is the coefficient of hydrogen, H_2 , when the following equation is balanced?



- A) 1
- B) 2
- C) 3
- D) 4
- E) 5

Answer: C

7) In this reaction, what is the coefficient for calcium oxide? $CaO(s) + CO_2(g) \rightarrow$

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