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Kamil Mrzygłód

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AI Value Creators Rob Thomas,Paul Zikopoulos,Kate Soule,2025-04 We've arrived in a new era GenAI is reshaping industries and decision making processes across the board As a result understanding their potential and pitfalls has become crucial But in order to stay ahead of the curve you'll need to develop fresh perspectives on leveraging AI beyond mere technical know-how Geared toward business leaders and tech professionals alike this book demystifies the strategic integration of AI into business practices ensuring you're equipped not just to participate but to lead in this new landscape This insightful guide by industry leaders Rob Thomas Paul Zikopoulos and Kate Soule goes beyond the basics offering real life success stories and learned lessons to provide a blueprint for meaningful AI engagement Whether you're a novice or a seasoned expert you'll come away with an enhanced understanding of GenAI Recognize the transformative potential of AI in business and how to harness it Navigate the ethical and operational challenges posed by AI with confidence Understand the dynamic interplay between AI technology and business strategy Implement actionable strategies to integrate AI into your organizational culture Step confidently into the role of an AI value creator equipped to lead and innovate

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Apps Azure Container Registry ACR and App Service Design powerful workflows with both low code and full code approaches Enhance applications with AI and machine learning components Who this book is for This book is for cloud developers and engineers building applications with Microsoft Azure as well as those looking to begin a career in Azure development While a basic understanding of programming concepts is recommended the book covers both basic and advanced ideas and solutions making it valuable for beginners and experienced developers looking to enhance their skills

Building AI Applications with Microsoft Semantic Kernel Lucas A. Meyer, 2024-06-21 Unlock the power of GenAI by effortlessly linking your C and Python apps with cutting edge models orchestrating diverse AI services with finesse and crafting bespoke applications through immersive real world examples Key Features Link your C and Python applications with the latest AI models from OpenAI Combine and orchestrate different AI services such as text and image generators Create your own AI apps with real world use case examples that show you how to use basic generative AI create images process documents use a vector database Purchase of the print or Kindle book includes a free PDF eBook Book Description In the fast paced world of AI developers are constantly seeking efficient ways to integrate AI capabilities into their apps Microsoft Semantic Kernel simplifies this process by using the GenAI features from Microsoft and OpenAI Written by Lucas A Meyer a Principal Research Scientist in Microsoft's AI for Good Lab this book helps you get hands on with Semantic Kernel It begins by introducing you to different generative AI services such as GPT 3.5 and GPT 4 demonstrating their integration with Semantic Kernel You'll then learn to craft prompt templates for reuse across various AI services and variables Next you'll learn how to add functionality to Semantic Kernel by creating your own plugins The second part of the book shows you how to combine multiple plugins to execute complex actions and how to let Semantic Kernel use its own AI to solve complex problems by calling plugins including the ones made by you The book concludes by teaching you how to use vector databases to expand the memory of your AI services and how to help AI remember the context of earlier requests You'll also be guided through several real world examples of applications such as RAG and custom GPT agents By the end of this book you'll have gained the knowledge you need to start using Semantic Kernel to add AI capabilities to your applications What you will learn Write reusable AI prompts and connect to different AI providers Create new plugins that extend the capabilities of AI services Understand how to combine multiple plugins to execute complex actions Orchestrate multiple AI services to accomplish a task Leverage the powerful planner to automatically create appropriate AI calls Use vector databases as additional memory for your AI tasks Deploy your application to ChatGPT making it available to hundreds of millions of users Who this book is for This book is for beginner level to experienced .NET or Python software developers who want to quickly incorporate the latest AI technologies into their applications without having to learn the details of every new AI service Product managers with some development experience will find this book helpful while creating proof of concept applications This book requires working knowledge of programming basics

Advances in Databases and Information Systems Joe Tekli, Johann

Gamper, Richard Chbeir, Yannis Manolopoulos, 2024-09-01 This volume LNCS 14918 constitutes the refereed proceedings of 28th European Conference ADBIS 2024 held in Bayonne France during August 28 31 2024 The 15 full papers presented were carefully reviewed and selected from 43 submissions The conference focuses on Algebra Models Schemata Discovery and Data Analysis Algorithms and Optimization Access Methods and Query Processing Advanced Architectures Machine Learning Large Language Models

The Singularity Is Nearer Ray Kurzweil, 2024-06-25 AN INSTANT NEW YORK TIMES BESTSELLER ONE OF TIME'S 100 MOST INFLUENTIAL PEOPLE IN ARTIFICIAL INTELLIGENCE The noted inventor and futurist's successor to his landmark book *The Singularity Is Near* explores how technology will transform the human race in the decades to come Since it was first published in 2005 Ray Kurzweil's *The Singularity Is Near* and its vision of an exponential future have spawned a worldwide movement Kurzweil's predictions about technological advancements have largely come true with concepts like AI intelligent machines and biotechnology now widely familiar to the public In this entirely new book Ray Kurzweil brings a fresh perspective to advances toward the Singularity assessing his 1999 prediction that AI will reach human level intelligence by 2029 and examining the exponential growth of technology that in the near future will expand human intelligence a millionfold and change human life forever Among the topics he discusses are rebuilding the world atom by atom with devices like nanobots radical life extension beyond the current age limit of 120 reinventing intelligence by connecting our brains to the cloud how exponential technologies are propelling innovation forward in all industries and improving all aspects of our well being such as declining poverty and violence and the growth of renewable energy and 3D printing He also considers the potential perils of biotechnology nanotechnology and artificial intelligence including such topics of current controversy as how AI will impact employment and the safety of autonomous cars and After Life technology which aims to virtually revive deceased individuals through a combination of their data and DNA The culmination of six decades of research on artificial intelligence *The Singularity Is Nearer* is Ray Kurzweil's crowning contribution to the story of this science and the revolution that is to come

Silicon Empires Nick Srnicek, 2025-12-23 Since the emergence of ChatGPT generative AI has been heralded as a technology poised to revolutionize our world But beyond the hype and hyperbole who truly wields power over this transformative technology In *Silicon Empires* Nick Srnicek explores the geopolitical economy of artificial intelligence revealing how a handful of powerful corporations and states are engaged in a monumental struggle to control its future Srnicek moves beyond the headlines to lay bare the elaborate strategies that these silicon empires from tech giants to great powers are deploying to capture the immense value of AI This incisive analysis uncovers the deep seated tensions between corporate ambitions and national interests and the profound consequences of this new era of technological competition As the race for AI supremacy accelerates Srnicek compellingly demonstrates that the decisions being made in the boardrooms of Silicon Valley and the halls of government will shape the distribution of wealth and power on a global scale for decades to come

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web jan 26 2016 the solubility product or ksp is a way of showing how effective a chemical is at dissolving into an aqueous solution the higher the ksp the more the chemical will dissolve into that solvent for this lab calcium hydroxide is created then ionized Ca(OH)_2 when calcium is put in water reacts as follows $\text{Ca(OH)}_2 \rightleftharpoons \text{Ca}^{2+} + 2\text{OH}^-$

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web k sp of calcium hydroxide 1 author j m mccormick last update ocober 13 2013 introduction an equilibrium constant k is related to ΔG for a process through eqn 1 and as long as ΔH and ΔS are independent of temperature eqn 1 can be combined with eqn 2 to determine ΔH and ΔS from the temperature dependence of k

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web sep 18 2020 in this experiment we will titrate a saturated calcium hydroxide solution with a standardized hcl solution we will determine the hydroxide ion concentration for the saturated calcium hydroxide

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web your primary objective in this experiment is to test a saturated solution of calcium hydroxide and use your observations and measurements to calculate the ksp of the compound you will do this by titrating the prepared Ca(OH)_2 solution with a standard hydrochloric acid solution

chem 114 exp 9 mike university of rhode island - May 04 2023

web in today s experiment you will determine the solubility product ksp of calcium hydroxide Ca(OH)_2 by measuring the concentration of Ca(OH)_2 in a saturated solution calcium hydroxide is a sparingly soluble salt that dissolves according to the following reaction the solubility product expression for this reaction is

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web expert verified chem 182 determining the ksp of calcium hydroxide calcium hydroxide is an ionic solid that is sparingly soluble in water a saturated aqueous solution of Ca(OH)_2 is represented in equation form as shown below $\text{Ca(OH)}_2 \rightleftharpoons \text{Ca}^{2+} + 2\text{OH}^-$ the solubility product expression describes in mathematical terms the equilibrium

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web ksp for calcium hydroxide calcium hydroxide is a soft white caustic powder used in making mortar cements calcium salts paints and petrochemicals it is also used in saltwater aquaria to make up kalkwasser limewater solutions for reef tanks and is used as a ph regulating agent notice that calcium hydroxide is divalent and thus has twice

ksp calcium hydroxide chem132 determination of the ksp of calcium - Jun 24 2022

web use the concentration of calcium hydroxide to calculate the concentration of calcium ions and hydroxide ions in the solution $\text{Ca(OH)}_2 \rightleftharpoons \text{Ca}^{2+} + 2\text{OH}^-$ $5 \times 10^{-2} \text{ M}$ calcium ions $5 \times 10^{-2} \text{ M}$ $2 \times 10^{-2} \text{ M}$ hydroxide ions calculate calcium hydroxide ksp using the concentrations of calcium ions and hydroxide ions

solubility of calcium hydroxide wpmu dev - Oct 29 2022

web madison bowe elana keller april 15 2014 introduction the objective of this lab was to measure the ksp of a slightly soluble basic salt and how solubility changes with temperature ksp is another form of equilibrium expression that deals with solubility the reaction that will take place is $\text{Ca(OH)}_2 \rightleftharpoons \text{Ca}^{2+} + 2\text{OH}^-$

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web question experiment 8 determining ksp for calcium hydroxide post lab report sheet data table 2 points trial 1 trial 2 hci molarity 0.050 M volume of filtered Ca(OH)_2 solution ml 5.00 5.00 3.71 13.75 equivalence point volume ml oh in saturated Ca(OH)_2 M Ca^{2+} in saturated Ca(OH)_2 M ksp of Ca(OH)_2 average ksp of Ca(OH)_2

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web the chemical equation used was $\text{Ca(OH)}_2 \rightleftharpoons \text{Ca}^{2+} + 2\text{OH}^-$ the solubility product was determined to be ksp Ca(OH)_2 from the titration which was made by dripping a concentrated solution of calcium hydroxide into hydrochloric acid the calcium hydroxide was filtered before starting the titration

18.1 solubility product constant ksp chemistry libretexts - Feb 01 2023

web jul 12 2023 because the concentration of a pure solid such as Ca_3PO_4 is a constant it does not appear explicitly in

the equilibrium constant expression the equilibrium constant expression for the dissolution of calcium phosphate is therefore

$$\text{Ca}_3(\text{PO}_4)_2 \rightleftharpoons 3\text{Ca}^{2+} + 2\text{PO}_4^{3-}$$

$$K_{\text{sp}} = [\text{Ca}^{2+}]^3 [\text{PO}_4^{3-}]^2$$

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web lab report 8 determinin of calcium hydroxide the purpose of this was to titrat e a saturated $\text{Ca}(\text{OH})_2$ soultion with a st andard hcl solution and find the ksp or the oh concentration was f ound and then the concentration was found and u for this experiment a titration of calcium hydroxide with hydrochlo

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web question determining the ksp of calciumhydroxide $\text{Ca}(\text{OH})_2$ s Ca^{2+} aq 2OH^- aq for this experiment the titration of calciumhydroxide with hydrochloric acid wasinvestigated determining the ksp of calcium hydroxide $\text{Ca}(\text{OH})_2$ s Ca^{2+} aq 2OH^- aq for this experiment the titration of calcium

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web question chem 182 experiment 8 determining the ksp of calcium hydroxide calcium hydroxide is an ionic solid that is sparingly soluble in water solution of $\text{Ca}(\text{OH})_2$ is represented in equation form lution of $\text{Ca}(\text{OH})_2$ s nronic solid that is sparingly soluble in water a saturated aqueous as shown below

determination of the ksp of calcium hydroxide autosaved - Jun 05 2023

web therefore it was determined through error calculations and ease of use that the most reliable valid method to use to get the closest ksp value to the theoretical ksp for $\text{Ca}(\text{OH})_2$ was the determination through ph method 2 introduction calcium hydroxide is a strong base soluble in water

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web we will filter the solid calcium hydroxide from the solution and then titrate the dissolved calcium hydroxide in solution with a known concentration of hydrochloric acid using titrimetric calculations we will determine the Ca^{2+} and OH^- concentrations in solution

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web calculate the ksp for calcium hydroxide Ca(OH)_2 $K_{sp} = 5 \times 10^{-6}$ find the accepted value of the ksp for calcium hydroxide and compare it with your value discuss the discrepancy and suggest possible sources of experimental error
accepted value for ksp for sodium hydroxide is 5×10^{-6}

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