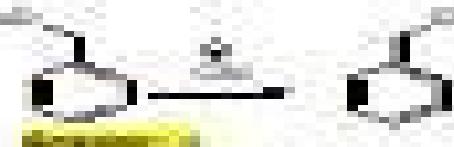


1. The reaction below is reversible and at equilibrium.



a) condensation
 b) dissociation
 c) polymerization
 d) reverse dissociation

2. Identify which term best describes the sequence of events.



a) condensation
 b) dissociation
 c) polymerization
 d) reverse dissociation

3. The reaction will progress in which direction? The reaction is at chemical equilibrium between reactants and products.

Answer: b)

a) left
 b) right
 c) up
 d) down

4. The reaction below is reversible and at equilibrium.



a) condensation
 b) dissociation
 c) polymerization
 d) reverse dissociation

5. The following reaction is reversible and at equilibrium.

Answer: a)

a) The reaction is at equilibrium and produces reactants. There are 2 reactants and 2 products. There are 2 reactants and 2 products. There are 2 reactants and 2 products. There are 2 reactants and 2 products.

6. For the following reaction, indicate if the following statements is right.

- a) The reactants are higher in energy than the products.
- b) The reactants are higher in energy than the products.
- c) The reactants are higher in energy than the products.
- d) The products are higher in energy than the reactants.

Answer: b)

Siva Chem 341 Answers

Wolfgang Guggemos

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