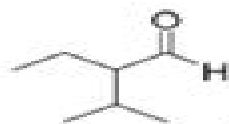


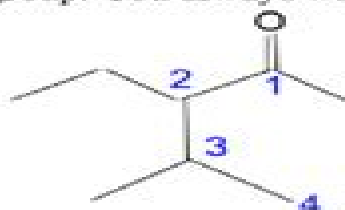
1. What is the correct IUPAC name of the following compound?



- a. 2-isopropylbutanal
- b. 2-ethyl-3-methylbutanal
- c. 2-isopropylbutanone
- d. 3-ethyl-2-methylbutanal

Correct Answer: B

Explanation: When naming aldehydes, you always want to change the -e from the parent name to -al. For example, hexane will be switched to hexanal if an aldehyde is present. We can already eliminate c, because we are not dealing with any ketones as our functional group. You always want to start where the carbonyl double bond is, so



that would be 1.

Now that the compound is numbered look at the answer choices. A cannot be the answer because there is no isopropyl group on carbon 2. Also, it's missing the other substituent of carbon 3. Answer choice B, makes the most sense, because answer choice D would mean that the carbonyl double bond would not be given the smallest number. You can never assign the larger number to the aldehyde, ketone, double bond, or triple bond; always try to give it the smallest number.

2. Rank the compounds in order of decreasing reactivity with LiAlH_4

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Thirumalaisamy P.
Velavancorresponding



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