

Balancing Equations Worksheet

- 1) $\underline{\cancel{1}} \text{Na}_3\text{PO}_4 + \underline{\cancel{3}} \text{KOH} \rightarrow \underline{\cancel{3}} \text{NaOH} + \underline{\cancel{1}} \text{K}_3\text{PO}_4$ Na-3, P-1, O-4, K-1
- 2) $\underline{\cancel{1}} \text{MgF}_2 + \underline{\cancel{1}} \text{Li}_2\text{CO}_3 \rightarrow \underline{\cancel{1}} \text{MgCO}_3 + \underline{2} \text{LiF}$ Mg-1, F-2, Li-2, C-1, O-3
- 3) $\underline{\cancel{1}} \text{P}_4 + \underline{3} \text{O}_2 \rightarrow \underline{1} \text{P}_2\text{O}_3$ P-4, O-2, P-2, O-3
- 4) $\underline{2} \text{RbNO}_3 + \underline{\cancel{1}} \text{BeF}_2 \rightarrow \underline{\cancel{1}} \text{Be(NO}_3)_2 + \underline{2} \text{RbF}$ Rb-1, N-1, O-3, Be-1, F-2
- 5) $\underline{2} \text{AgNO}_3 + \underline{\cancel{1}} \text{Cu} \rightarrow \underline{\cancel{1}} \text{Cu(NO}_3)_2 + \underline{2} \text{Ag}$ Ag-1, N-1, O-3, Cu-1
- 6) $\underline{\cancel{1}} \text{CF}_4 + \underline{2} \text{Br}_2 \rightarrow \underline{\cancel{1}} \text{CBr}_4 + \underline{2} \text{F}_2$ C-1, F-4, Br-2, C-1, Br-4, F-2
- 7) $\underline{2} \text{HCN} + \underline{\cancel{1}} \text{CuSO}_4 \rightarrow \underline{\cancel{1}} \text{H}_2\text{SO}_4 + \underline{\cancel{1}} \text{Cu(CN)}_2$ H-1, C-1, N-1, Cu-1, S-1, O-4
- 8) $\underline{\cancel{1}} \text{GaF}_3 + \underline{3} \text{Cs} \rightarrow \underline{3} \text{CsF} + \underline{\cancel{1}} \text{Ga}$ Ga-1, F-3, Cs-1, Ga-1, F-3
- 9) $\underline{\cancel{1}} \text{BaS} + \underline{\cancel{1}} \text{PtF}_2 \rightarrow \underline{\cancel{1}} \text{BaF}_2 + \underline{\cancel{1}} \text{PtS}$ Ba-1, S-1, Pt-1, F-2, Ba-1, F-2, Pt-1, S-1
- 10) $\underline{\cancel{1}} \text{N}_2 + \underline{3} \text{H}_2 \rightarrow \underline{2} \text{NH}_3$ N-2, H-2, N-1, H-3
- 11) $\underline{2} \text{NaF} + \underline{\cancel{1}} \text{Br}_2 \rightarrow \underline{2} \text{NaBr} + \underline{\cancel{1}} \text{F}_2$ Na-1, F-1, Br-2, Na-2, Br-2, F-2
- 12) $\underline{\cancel{1}} \text{Pb(OH)}_2 + \underline{2} \text{HCl} \rightarrow \underline{2} \text{H}_2\text{O} + \underline{\cancel{1}} \text{PbCl}_2$ Pb-1, O-2, H-2, Cl-1, Pb-1, O-2, H-2, Cl-2
- 13) $\underline{2} \text{AlBr}_3 + \underline{3} \text{K}_2\text{SO}_4 \rightarrow \underline{6} \text{KBr} + \underline{\cancel{1}} \text{Al}_2(\text{SO}_4)_3$ Al-2, Br-6, K-2, S-1, O-4, Al-2, Br-6, K-6, S-3, O-12
- 14) $\underline{1} \text{CH}_4 + \underline{2} \text{O}_2 \rightarrow \underline{1} \text{CO}_2 + \underline{2} \text{H}_2\text{O}$ C-1, H-4, O-2, C-1, O-2, H-2, O-1
- 15) $\underline{2} \text{Na}_3\text{PO}_4 + \underline{3} \text{CaCl}_2 \rightarrow \underline{3} \text{NaCl} + \underline{\cancel{1}} \text{Ca}_3(\text{PO}_4)_2$ Na-3, P-1, O-4, Ca-1, Cl-2, Na-3, P-1, O-4, Ca-3, P-2, O-8
- 16) $\underline{2} \text{K} + \underline{\cancel{1}} \text{Cl}_2 \rightarrow \underline{2} \text{KCl}$ K-1, Cl-2, K-2, Cl-2
- 17) $\underline{2} \text{Al} + \underline{6} \text{HCl} \rightarrow \underline{3} \text{H}_2 + \underline{2} \text{AlCl}_3$ Al-1, H-1, Cl-1, Al-2, H-6, Cl-6, H-3, Cl-6
- 18) $\underline{\cancel{1}} \text{N}_2 + \underline{3} \text{F}_2 \rightarrow \underline{2} \text{NF}_3$ N-2, F-2, N-1, F-6
- 19) $\underline{\cancel{1}} \text{SO}_2 + \underline{2} \text{Li}_2\text{Se} \rightarrow \underline{\cancel{1}} \text{SSe}_2 + \underline{2} \text{Li}_2\text{O}$ S-1, O-2, Li-2, Se-1, S-1, Se-2, Li-2, O-2
- 20) $\underline{2} \text{NH}_3 + \underline{\cancel{1}} \text{H}_2\text{SO}_4 \rightarrow \underline{\cancel{1}} (\text{NH}_4)_2\text{SO}_4$ N-1, H-3, S-1, O-4, N-2, H-8, S-1, O-4

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Michael Brown



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