

## SALT ANALYSIS

### TEST FOR ANIONS

EXPERIMENT	OBSERVATION	INFERENCE
1. Test for $\text{CO}_3^{2-}$ → A little of the salt is treated with dil. HCl.	Brisk effervescence with the evolution of colourless odourless gas	May be $\text{CO}_3^{2-}$ $\text{CaCO}_3 + 2\text{HCl} \rightarrow \text{CaCl}_2 + \text{CO}_2 + \text{H}_2\text{O}$
Confirmatory test :- 2. Pass the above gas through lime water.	Lime water turns milky	Presence of $\text{CO}_3^{2-}$ is confirmed $\text{Ca(OH)}_2 + \text{CO}_2 \rightarrow \text{CaCO}_3 + \text{H}_2\text{O}$
Test for Halides 3. A little of the salt is treated with 1 or 2 drops of conc. $\text{H}_2\text{SO}_4$	a) Colourless pungent smelling gas evolves, dense white fumes when a glass rod is dipped in $\text{NH}_4\text{OH}$ is shown on the mouth of the test tube b) Reddish brown vapours with pungent smell  c) Deep violet vapours with pungent smell	a) May be $\text{Cl}^-$ $\text{NaCl} + \text{H}_2\text{SO}_4 \rightarrow \text{NaHSO}_4 + \text{HCl}$  b) Maybe $\text{Br}^-$ $\text{NaBr} + \text{H}_2\text{SO}_4 \rightarrow \text{NaHSO}_4 + \text{HBr}$ $\text{H}_2\text{SO}_4 + 2\text{HBr} \rightarrow \text{SO}_2 + \text{Br}_2 + 2\text{H}_2\text{O}$  c) May be $\text{I}^-$ $\text{KI} + \text{H}_2\text{SO}_4 \rightarrow \text{KHSO}_4 + \text{HI}$ $\text{H}_2\text{SO}_4 + 2\text{HI} \rightarrow \text{SO}_2 + \text{I}_2 + 2\text{H}_2\text{O}$
Confirmatory test of halides 4. A little of the salt solution is acidified with dil. $\text{HNO}_3$ , boil, cool and add $\text{AgNO}_3$ sol.	a) Curdy white ppt. soluble in $\text{NH}_4\text{OH}$ .  b) Pale yellow ppt. sparingly soluble in $\text{NH}_4\text{OH}$  c) Yellow ppt. insoluble in $\text{NH}_4\text{OH}$	a) presence of $\text{Cl}^-$ is confirmed $\text{NaCl} + \text{AgNO}_3 \rightarrow \text{NaNO}_3 + \text{AgCl (white ppt.)}$  b) presence of $\text{Br}^-$ is confirmed $\text{KBr} + \text{AgNO}_3 \rightarrow \text{KNO}_3 + \text{AgBr (pale yellow ppt.)}$  c) presence of $\text{I}^-$ is confirmed $\text{KI} + \text{AgNO}_3 \rightarrow \text{KNO}_3 + \text{AgI (yellow ppt.)}$

# Unknown Salt Identification

**David Collins (Ph. D.), Matthew Johll**



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