

1. The reaction below is an example of:



a combination
 a decomposition
 a synthesis
 a double displacement

2. Identify which term best describes the sequence of:



a combination
 a decomposition
 a synthesis
 a sequence

3. The reaction will progress in which direction? The reaction is between potassium bromide and silver nitrate.

Answer: to the right

left
 right
 up
 down

4. The reaction below is an example of:



a combination
 a decomposition
 a synthesis
 a double displacement

5. The following reaction is an example of:



the combination of two reactants to form a single product. This reaction is called a synthesis reaction. It is also called a condensation reaction because the reactants are joined together to form a single product. Some examples of synthesis reactions are:

6. The following reaction is an example of the following reaction because:

- The reactants are higher in energy than the products.
- The reactants are higher in energy than the products.
- The reactants are higher in energy than the products.
- The products are higher in energy than the reactants.

Answer: C

Siva Chem 341 Answers

Dr. Viraf J. Dalal

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