

# Molarity Practice

1. Calculate the molarity of the following solutions:
  - a. 10.0 g of NaOH in 250 mL of solution
  - b. 10.0 g of sucrose,  $C_{12}H_{22}O_{11}$ , in 250 mL of solution
2. Calculate the molarity of the following solutions:
  - a. 0.500 g of NaCl in 250 mL of solution
  - b. 0.500 g of sucrose,  $C_{12}H_{22}O_{11}$ , in 250 mL of solution
3. Calculate the molarity of the following solutions:
  - a. 0.500 g of sucrose and 0.500 g of NaCl in 250 mL of solution
  - b. 0.500 g of sucrose and 0.500 g of NaCl in 500 mL of solution
4. Calculate the molarity of the following solutions for the following concentrations:  $1.00 \times 10^{-3} M$ 
  - a. 0.010 g of NaOH in 250 mL of solution
  - b. 0.010 g of NaCl in 250 mL of solution
5. Calculate the molarity of the following solutions:
  - a. 0.001 g of NaCl in 250 mL of solution
  - b. 0.001 g of sucrose in 250 mL of solution

## Skill Practice 53 Answers Chemistry

**California,Carter Pitkin Pomeroy**

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**Student Solutions Manual** Darrell Ebbing,2005-03-17 Provides solutions to odd numbered Practice Problems General Problems and Cumulative Skills Problems plus answers to Review Questions Ebbing Darrell D. Ebbing,1993 **Catalog**

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