

Molarity Practice

1. Calculate the molarity of the following solutions:

- 10.0 g of NaCl in 250 mL of solution

2. 10.0 g of sucrose, C₁₂H₂₂O₁₁, in 250 mL of solution

3. 10.0 g of NaClO₄ in 250 mL of solution

4. 10.0 g of sucrose in 250 mL of solution

5. 10.0 g of sucrose and 10.0 g of NaCl in 250 mL of solution

6. 10.0 g of sucrose and 10.0 g of NaCl in 250 mL of solution

7. 10.0 g of sucrose and 10.0 g of NaCl in 250 mL of solution

8. Calculate the molarity of the following solutions:

- 10.0 g of NaCl in 250 mL of solution
- 10.0 g of sucrose in 250 mL of solution

9. Calculate the molarity of the following solutions:

- 10.0 g of sucrose and 10.0 g of NaCl in 250 mL of solution

Skill Practice 53 Answers Chemistry

California,Carter Pitkin Pomeroy

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Student Solutions Manual Darrell Ebbing,2005-03-17 Provides solutions to odd numbered Practice Problems General Problems and Cumulative Skills Problems plus answers to Review Questions Ebbing Darrell D. Ebbing,1993 **Catalog**

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