

Name \_\_\_\_\_ Date \_\_\_\_\_ Class \_\_\_\_\_

## CHAPTER 15 STUDY GUIDE FOR CONTENT MASTERY

### Solutions

#### Section 15.1 What are solutions?

In your textbook, read about the characteristics of solutions.

Use each of the terms below just once to complete the passage.

immiscible	liquid	solution
insoluble	miscible	soluble
		solvent

Air is air! (1) solution of oxygen gas dissolved in nitrogen gas. The oxygen is air is the (2) soluble and nitrogen is the (3) solvent. Because oxygen gas dissolves in a solvent, oxygen gas is air is (4) soluble substances. A substance that does not dissolve is (5) insoluble. (6) liquid solutions are the most common type of solutions. If one liquid is soluble in another liquid, such as acetic acid in water, the two liquids are (7) miscible. However, if one liquid is insoluble in another, the liquids are (8) immiscible.

Read about hydration in aqueous solutions in your textbook.

The diagram shows the hydration of solid sodium chloride to form an aqueous solution. Use the diagram to answer the following questions.



8. Hydration is a solution in which the solvent is water. What is solute?

Solution is the process of surrounding solute particles with solvent particles to form a solution.

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## CHAPTER 15 STUDY GUIDE FOR CONTENT MASTERY

#### Section 15.1 continued

10. As sodium chloride dissolves in water, what happens to the sodium and chloride ions?

The sodium and chloride ions are separated and surrounded by the water molecules.

11. Explain the attraction of the water molecules around the sodium ions and chloride ions.

Because the sodium ion is positively charged, it attracts the negatively charged portion of the water molecule (the oxygen atom) and repels the positively charged portion of the water molecule (the hydrogen atoms). Because the chloride ion is negatively charged, it attracts the positively charged portion of the water molecule and repels the negatively charged portion.

12. How does the strength of the attraction between water molecules and sodium and chloride ions compare with the strength of the attraction between the sodium ions and chloride ions? How do you know?

The attraction between the water molecules and the sodium and chloride ions is greater than the attraction between the sodium and chloride ions. The greater strength of attraction between the water molecules and the ions is what causes the solution process to occur.

13. List three ways that the rate of solution may be increased.

stirring or shaking the solution, breaking the solute into smaller pieces, and heating the solvent

In your textbook, read about heat of solution, solubility, and factors that affect solubility.

For each statement below, write true or false.

True \_\_\_\_\_ 14. The overall energy change that occurs when a solution forms is called the heat of solution.

False \_\_\_\_\_ 15. Solubility is a measure of the maximum amount of solute that dissolves in a given amount of solvent at a specified temperature and pressure.

False \_\_\_\_\_ 16. Solvation continues as long as the solvation rate is less than the crystallization rate.

True \_\_\_\_\_ 17. In a saturated solution, solvation and crystallization are in equilibrium.

True \_\_\_\_\_ 18. Additional solute can be dissolved in an unsaturated solution.

True \_\_\_\_\_ 19. The solubility of a gas dissolved in a liquid decreases as the temperature of the solution increases.

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# Study Guide For Content Mastery

**William N. Bender**



## **Study Guide For Content Mastery:**

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**Study Guide to DSM-5-TR®** Laura Weiss Roberts, Alan K. Louie, Matthew L. Edwards, 2024-12-12 The ultimate companion volume to DSM 5 TR the Study Guide is designed to help clinical learners teachers and practitioners in psychiatry psychology and social work understand and apply diagnostic criteria Medical students residents and fellows and trainees throughout the clinical neurosciences will find this text to be invaluable in developing their diagnostic abilities Readers learn key clinical content through concise chapters and can assess their knowledge with more than 100 multiple choice questions The book is organized into three sections Part I delves into foundational concepts of psychiatric diagnoses and explores how diagnostic assessment fits into the process of working therapeutically with patients In this new edition Part I includes a new chapter examining structural and cultural considerations in arriving at a diagnosis Part II focuses on the diagnostic classes within DSM 5 TR and features clinical case vignettes with real life complexity relevant to psychiatric diagnoses diagnostic pearls that can be applied to daily clinical practice and a self assessment section with key concepts questions for further discussion patient cases and Q As Part III includes more than 100 multiple choice questions with an answer key These questions cover a range of diagnoses and patient presentations from across DSM 5 TR helping readers review their learning and assess their skills Taking a patient centered approach that complements the more disorder centered organization of DSM 5 TR this Study Guide provides context for DSM 5 TR s diagnoses while offering an array of interesting and memorable cases Rich in thought provoking detail and designed for repeated use this is an indispensable guide for learners educators and clinicians

**Student Study Guide With IBM® SPSS® Workbook for Research Methods for the Behavioral Sciences** Gregory J. Privitera, 2019-01-24 The Student Study Guide With IBM SPSS Workbook for Research Methods for the Behavioral Sciences Third Edition by Gregory J Privitera includes a review of chapter learning objectives chapter summaries and tips and cautions To help students practice their skills the guide offers quizzes and exercises accompanied by answers keys SPSS in Focus exercises with general instructions complement those in Privitera s main text Research Methods for the Behavioral Sciences Third Edition *Chemistry*, 2001-04-01 *Student Study Guide With IBM® SPSS® Workbook for Essential Statistics for the Behavioral Sciences* Gregory J. Privitera, 2015-01-06 The Student Study Guide With IBM SPSS Workbook for Essential Statistics for the Behavioral Sciences by Gregory J Privitera includes a review of chapter learning objectives chapter outlines and key terms essential statistical formulas special tips and insights for students and chapter summaries To help students practice skills the guide offers word searches and crossword puzzles for each chapter extensive

practice quizzes linked to chapter learning objectives and SPSS in Focus exercises which complement those in the book

**Glencoe Science Voyages** Alton Biggs, 2001 CD ROM Create interactive science voyages and conduct experiments Includes quizzes

**Differentiating Instruction for Students With Learning Disabilities** William N. Bender, 2008 Written for teachers who want to know how to differentiate instruction for students with learning difficulties the second edition of *Differentiating Instruction for Students with Learning Disabilities* draws upon the bestselling success of the first edition in explicitly showing what differentiated instruction is and how to use differentiated strategies in the classroom With more concentration on brain research multiple intelligence response to intervention RTI tiered instruction and universal design for learning UDL William N Bender provides the most comprehensive book on differentiated instruction and learning disabilities based on scientific research and with a practical teacher friendly approach An updated reference section brief Teaching Tips side bars and thought provoking new feature Reflections augment this powerful resource that educators can immediately use in the classroom with all students who have learning disabilities who are at risk and who may have learning difficulties

**Science Voyages** Glencoe, 1999-05 Chemistry: Matter & Change, Solving Problems - A Chemistry Handbook McGraw Hill, 2001-08 Glencoe Chemistry Solving Problems A Chemistry Handbook Matter and Change *Chemistry* McGraw-Hill Staff, 2001-07

**Resources in Education**, 1996-05 Student Study Guide Sheila R. Alber, David F. Bicard, 2002-07-15 *A Guidebook for Serving Students with Disabilities in Tech Prep*, 1996 A comprehensive guide to developing or revising vocational education programs for students with special needs including strategies guidelines and checklists

**Study Guide to Accompany Introduction to Critical Care Nursing** Dawn Ellen Zimmerman, Mary Lou Sole, Marilyn L. Lamborn, Jeanette Hartshorn, 2000-11 Concept maps for each chapter in the parent text help your students sharpen their critical thinking skills and numerous multiple choice review questions challenge their knowledge of each chapter's content Additional exercises such as crossword puzzles also facilitate learning and key point summaries bring it all together

Helping Adults Learn Alan B. Knox, 1986-10-24

**A Guide for Developing an English Curriculum for the Eighties** Allan A. Glatthorn, 1980 Intended for use by those who have direct responsibility for providing leadership in the area of English this monograph provides a process for improving the English curriculum in a way that is true to the intellectual and emotional needs of students while meeting society's expectations for a practical curriculum The monograph begins with a retrospective examination of the teaching of English follows with an analysis of present trends and sets forth a process of incremental curriculum development The remaining chapters provide a framework for a mastery curriculum that requires careful structuring Topics discussed include 1 curriculum mapping 2 building a content planning matrix 3 evaluating and modifying the matrix 4 developing the syncretic English curriculum 5 providing for mandated competencies 6 using research to improve the teaching of English 7 developing the English notebook 8 using separate objectives in integrated units 9 building English mastery into interdisciplinary courses 10 designing sound elective programs

and 11 writing mastery learning units An epilogue contains a discussion of a curriculum of meaning HOD      **Basic Nursing**  
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