

Introduction to Chemistry

Section 1.1 The Stories of Two Chemicals

In your textbook, read about the ozone layer.

Use each of the boxes below just once to complete the passage.

atmosphere	oxygen gas	ozone	ionized state
ionosphere	hydrogen	ultraviolet radiation	

Earth's **OZ** _____ is made up of several layers. The air ionosphere makes up the lowest level. This layer is called the **OZ** _____. The next layer up is called the **OZ** _____. This layer contains a protective _____.

UV _____ rays _____.

Chemical reactions **UV** _____ is caused by ultraviolet radiation in the upper part of the atmosphere. The ozone layer is being eroded slowly, without _____.

UV _____ Without ozone, you get more likely to get a rash from too much sunbathing. The thinning of the ozone layer called the **UV** _____ is considered because without ozone all organisms on Earth are subject to harm from too much radiation.

In your textbook, read about chlorofluorocarbons.

Use each statement below, with one or more _____.

- _____ **1.** CFCs are useful since they are chlorofluorocarbons.
- _____ **2.** CFCs are made up of carbon, fluorine, and chlorine.
- _____ **3.** All CFCs are synthetic chemicals.
- _____ **4.** CFCs readily react slowly with other chemicals.
- _____ **5.** CFCs were developed as replacements for toxic refrigerants.

Study Guide For Content Mastery Chapter 8

**Elizabeth M.H. Coghill, Jeffrey G.
Coghill**



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