

Name _____ Date _____ Class _____

CHAPTER 15 STUDY GUIDE FOR CONTENT MASTERY

Solutions

Section 15.1 What are solutions?

In your textbook, read about the characteristics of solutions.

Use each of the terms below just once to complete the passage.

immiscible	liquid	soluble	solution
insoluble	miscible	solvent	solvent

Air is air! (1) solution of oxygen gas dissolved in nitrogen gas. The oxygen in air is the (2) solute and nitrogen is the (3) solvent. Because oxygen gas dissolves in a solvent, oxygen gas is air! (4) soluble substances. A substance that does not dissolve is (5) insoluble. (6) liquid solutions are the most common type of solutions. If one liquid is soluble in another liquid, such as acetic acid in water, the two liquids are (7) miscible. However, if one liquid is insoluble in another, the liquids are (8) immiscible.

Read about solvation in aqueous solutions in your textbook.

The diagram shows the hydration of solid sodium chloride to form an aqueous solution. Use the diagram to answer the following questions.



8. Hydration is solvation in which the solvent is water. What is solvation?

Solvation is the process of surrounding solute particles with solvent particles to form a solution.

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Chemistry: Matter and Change • Chapter 15

85

Name _____ Date _____ Class _____

CHAPTER 15 STUDY GUIDE FOR CONTENT MASTERY

Section 15.1 continued

10. As sodium chloride dissolves in water, what happens to the sodium and chloride ions?

The sodium and chloride ions are separated and surrounded by the water molecules.

11. Explain the attraction of the water molecules around the sodium ions and chloride ions.

Because the sodium ion is positively charged, it attracts the negatively charged portion of the water molecule (the oxygen atom) and repels the positively charged portion of the water molecule (the hydrogen atoms). Because the chloride ion is negatively charged, it attracts the positively charged portion of the water molecule and repels the negatively charged portion.

12. How does the strength of the attraction between water molecules and sodium and chloride ions compare with the strength of the attraction between the sodium ions and chloride ions? How do you know?

The attraction between the water molecules and the sodium and chloride ions is greater than the attraction between the sodium and chloride ions. The greater strength of attraction between the water molecules and the ions is what causes the solvation process to occur.

13. List three ways that the rate of solvation may be increased.

stirring or shaking the solution, breaking the solute into smaller pieces, and heating the solvent

In your textbook, read about heat of solution, solubility, and factors that affect solubility.

For each statement below, write true or false.

True _____ 14. The overall energy change that occurs when a solution forms is called the heat of solution.

False _____ 15. Solubility is a measure of the maximum amount of solute that dissolves in a given amount of solvent at a specified temperature and pressure.

False _____ 16. Solvation continues as long as the solvation rate is less than the crystallization rate.

True _____ 17. In a saturated solution, solvation and crystallization are in equilibrium.

True _____ 18. Additional solute can be dissolved in an unsaturated solution.

True _____ 19. The solubility of a gas dissolved in a liquid decreases as the temperature of the solution increases.

86 Chemistry: Matter and Change • Chapter 15

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Alton Biggs



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Study Guide to DSM-5-TR® Laura Weiss Roberts, Alan K. Louie, Matthew L. Edwards, 2024-12-12 The ultimate companion volume to DSM 5 TR the Study Guide is designed to help clinical learners teachers and practitioners in psychiatry psychology and social work understand and apply diagnostic criteria Medical students residents and fellows and trainees throughout the clinical neurosciences will find this text to be invaluable in developing their diagnostic abilities Readers learn key clinical content through concise chapters and can assess their knowledge with more than 100 multiple choice questions The book is organized into three sections Part I delves into foundational concepts of psychiatric diagnoses and explores how diagnostic assessment fits into the process of working therapeutically with patients In this new edition Part I includes a new chapter examining structural and cultural considerations in arriving at a diagnosis Part II focuses on the diagnostic classes within DSM 5 TR and features clinical case vignettes with real life complexity relevant to psychiatric diagnoses diagnostic pearls that can be applied to daily clinical practice and a self assessment section with key concepts questions for further discussion patient cases and Q As Part III includes more than 100 multiple choice questions with an answer key These questions cover a range of diagnoses and patient presentations from across DSM 5 TR helping readers review their learning and assess their skills Taking a patient centered approach that complements the more disorder centered organization of DSM 5 TR this Study Guide provides context for DSM 5 TR s diagnoses while offering an array of interesting and memorable cases Rich in thought provoking detail and designed for repeated use this is an indispensable guide for learners educators and clinicians

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A Guide for Developing an English Curriculum for the Eighties Allan A. Glatthorn,1980 Intended for use by those who have direct responsibility for providing leadership in the area of English this monograph provides a process for improving the English curriculum in a way that is true to the intellectual and emotional needs of students while meeting society s expectations for a practical curriculum The monograph begins with a retrospective examination of the teaching of English follows with an analysis of present trends and sets forth a process of incremental curriculum development The remaining chapters provide a framework for a mastery curriculum that requires careful structuring Topics discussed include 1 curriculum mapping 2 building a content planning matrix 3 evaluating and modifying the matrix 4 developing the syncretic English curriculum 5 providing for mandated competencies 6 using research to improve the teaching of English 7 developing the English notebook 8 using separate objectives in integrated units 9 building English mastery into interdisciplinary courses 10 designing sound elective programs and 11 writing mastery learning units An epilogue contains a discussion of a curriculum of meaning HOD
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Table of Contents Study Guide For Content Mastory

1. Understanding the eBook Study Guide For Content Mastory
 - The Rise of Digital Reading Study Guide For Content Mastory
 - Advantages of eBooks Over Traditional Books
2. Identifying Study Guide For Content Mastory
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Study Guide For Content Mastory
 - User-Friendly Interface
4. Exploring eBook Recommendations from Study Guide For Content Mastory
 - Personalized Recommendations
 - Study Guide For Content Mastory User Reviews and Ratings
 - Study Guide For Content Mastory and Bestseller Lists
5. Accessing Study Guide For Content Mastory Free and Paid eBooks

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- Study Guide For Content Mastory eBook Subscription Services
- Study Guide For Content Mastory Budget-Friendly Options
- 6. Navigating Study Guide For Content Mastory eBook Formats
 - ePub, PDF, MOBI, and More
 - Study Guide For Content Mastory Compatibility with Devices
 - Study Guide For Content Mastory Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Study Guide For Content Mastory
 - Highlighting and Note-Taking Study Guide For Content Mastory
 - Interactive Elements Study Guide For Content Mastory
- 8. Staying Engaged with Study Guide For Content Mastory
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Study Guide For Content Mastory
- 9. Balancing eBooks and Physical Books Study Guide For Content Mastory
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Study Guide For Content Mastory
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Study Guide For Content Mastory
 - Setting Reading Goals Study Guide For Content Mastory
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Study Guide For Content Mastory
 - Fact-Checking eBook Content of Study Guide For Content Mastory
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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