



# CHEMISTRY

## CHEAT SHEET

### WRITING IONIC FORMULAS



#### IONIC COMPOUNDS CHEAT SHEET

##### WRITING IONIC FORMULAS

**How to Write Ionic Formulas**

1. Write the symbols for the cation and anion, including their charges.

2. Use the charges to determine the ratio of cations to anions needed to balance the charges.

3. Write the formula, using subscripts to indicate the number of each ion.

4. Check the formula to make sure the charges are balanced.

CATIONS		ANIONS	
1	$\text{Na}^+$	1	$\text{Cl}^-$
2	$\text{K}^+$	2	$\text{O}^{2-}$
3	$\text{Ca}^{2+}$	3	$\text{F}^-$
4	$\text{Mg}^{2+}$	4	$\text{Br}^-$
5	$\text{Al}^{3+}$	5	$\text{I}^-$
6	$\text{Fe}^{2+}$	6	$\text{S}^{2-}$
7	$\text{Fe}^{3+}$	7	$\text{N}^{3-}$
8	$\text{Zn}^{2+}$	8	$\text{P}^{3-}$
9	$\text{Ba}^{2+}$	9	$\text{As}^{3-}$
10	$\text{Sr}^{2+}$	10	$\text{Sb}^{3-}$
11	$\text{Pb}^{2+}$	11	$\text{Bi}^{3-}$
12	$\text{Pb}^{4+}$	12	$\text{Mn}^{2+}$
13	$\text{Sn}^{2+}$	13	$\text{Mn}^{4+}$
14	$\text{Sn}^{4+}$	14	$\text{Cr}^{3+}$
15	$\text{Co}^{2+}$	15	$\text{Cr}^{6+}$
16	$\text{Co}^{3+}$	16	$\text{Mg}^{2+}$
17	$\text{Ni}^{2+}$	17	$\text{Ni}^{4+}$
18	$\text{Ni}^{3+}$	18	$\text{Cu}^+$
19	$\text{Cu}^{2+}$	19	$\text{Cu}^{2+}$
20	$\text{Ag}^+$	20	$\text{Ag}^+$
21	$\text{Hg}^{2+}$	21	$\text{Hg}^{2+}$
22	$\text{Hg}^{2+}$	22	$\text{Hg}^{2+}$
23	$\text{Hg}^{2+}$	23	$\text{Hg}^{2+}$
24	$\text{Hg}^{2+}$	24	$\text{Hg}^{2+}$
25	$\text{Hg}^{2+}$	25	$\text{Hg}^{2+}$
26	$\text{Hg}^{2+}$	26	$\text{Hg}^{2+}$
27	$\text{Hg}^{2+}$	27	$\text{Hg}^{2+}$
28	$\text{Hg}^{2+}$	28	$\text{Hg}^{2+}$
29	$\text{Hg}^{2+}$	29	$\text{Hg}^{2+}$
30	$\text{Hg}^{2+}$	30	$\text{Hg}^{2+}$

**How to Write Molecular Formulas**

1. Write the symbols for the elements, including their valences.

2. Use the valences to determine the ratio of atoms needed to balance the valences.

3. Write the formula, using subscripts to indicate the number of each atom.

4. Check the formula to make sure the valences are balanced.

1	2	3	4
1	1	1	1
2	1	1	1
3	1	1	1

1	2	3	4
1	1	1	1
2	1	1	1
3	1	1	1
4	1	1	1
5	1	1	1
6	1	1	1
7	1	1	1
8	1	1	1
9	1	1	1
10	1	1	1
11	1	1	1
12	1	1	1
13	1	1	1
14	1	1	1
15	1	1	1
16	1	1	1
17	1	1	1
18	1	1	1
19	1	1	1
20	1	1	1
21	1	1	1
22	1	1	1
23	1	1	1
24	1	1	1
25	1	1	1
26	1	1	1
27	1	1	1
28	1	1	1
29	1	1	1
30	1	1	1

1	2
1	1
2	1
3	1
4	1
5	1
6	1
7	1
8	1
9	1
10	1
11	1
12	1
13	1
14	1
15	1
16	1
17	1
18	1
19	1
20	1
21	1
22	1
23	1
24	1
25	1
26	1
27	1
28	1
29	1
30	1

# Writing Ionic Formulas Study Guide

**Robinson**



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