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buffers typically consist of an acid base pair with the acid and base differing by the presence or absence of a proton a
 conjugate acid base pair for instance one of the buffers that maintain the ph of human blood involves carbonic acid H_2CO_3
 and its conjugate base the bicarbonate ion HCO_3^-

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the reaction of an acid and a base is called a neutralization reaction although acids and bases have their own unique chemistries the acid and base cancel each other's chemistry to produce a rather innocuous substance water in fact the general reaction between an acid and a base is

$$acid + base \rightarrow water + salt$$

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