

**CHAPTER 15** **STUDY GUIDE****Energy and Chemical Change****Section 15.1 Energy**

*In your textbook, read about the nature of energy.*

**In the space at the left, write *true* if the statement is true; if the statement is false, change the italicized word or phrase to make it true.**

1. Energy is the ability to do work or produce heat.
2. The law of conservation of energy states that energy *can be* created and destroyed.
3. Chemical potential energy is energy stored in a substance because of its *composition*.
4. *Heat* is a form of energy that flows from a warmer object to a cooler object.
5. A calorie is the amount of energy required to raise the temperature of *one gram* of pure water by one degree Celsius.
6. A *calorie* is the SI unit of heat and energy.
7. The *specific* heat of a substance is the amount of heat required to raise the temperature of one gram of that substance by one degree Celsius.
8. *Kinetic* energy is energy of motion.
9. Chemicals participating in a chemical reaction contain only *potential energy*.
10. One nutritional Calorie is equal to *100 calories*.
11. One calorie equals *4.184 joules*.
12. When a fuel is burned, some of its *chemical potential energy* is lost as heat.
13. To convert kilojoules to joules, *divide* the number of kilojoules by 1000 joules/1 kilojoule.

**Answer the following question. Show all your work.**

14. If the temperature of a 500.0-g sample of liquid water is raised 2.00°C, how much heat is absorbed by the water? The specific heat of liquid water is 4.184 J/(g·°C).

# Study Guide Energy Chemical Change Answer Key

**Herman Thompson Briscoe**

## **Study Guide Energy Chemical Change Answer Key:**

**Chemistry Terminology III (Speedy Study Guides)** Speedy Publishing, 2015-04-27 This third volume lists commonly confused Chemistry terminologies from P to V and will complete your special chemical dictionary Understanding terminologies is a prerequisite to solving chemical equations and understanding chemical relationships Be prepared to impress with A's on your next chemistry exam Don't forget to grab a copy of this third volume too

**Barron's Science**

**360: A Complete Study Guide to Chemistry with Online Practice** Barron's Educational Series, Mark Kernion, Joseph A. Mascetta, 2021-09-07 Barron's Science 360 Chemistry is your complete go to guide for everything chemistry This comprehensive guide is an essential resource for High school and college courses Homeschooling Virtual Learning Learning pods Inside you'll find Comprehensive Content Review Begin your study with the basic building block of chemistry and build as you go Topics include atomic structure chemical formulas electrochemistry the basics of organic chemistry and much more Effective Organization Topic organization and simple lesson formats break down the subject matter into manageable learning modules that help guide a successful study plan customized to your needs Clear Examples and Illustrations Easy to follow explanations hundreds of helpful illustrations and numerous step by step examples make this book ideal for self study and rapid learning Practice Exercises Each chapter ends with practice exercises designed to reinforce and extend key skills and concepts These checkup exercises along with the answers and solutions will help you assess your understanding and monitor your progress Access to Online Practice Take your learning online for 50 practice questions designed to test your knowledge with automated scoring to show you how far you have come

**Study Guide, Student Edition** Merrill, 1994-07

**Study Guide for Chemical Principles** Peter Atkins, John Krenos, Joseph Potenza, 2013-03-01 The Student Study Guide helps students to improve their problem solving skills avoid common mistakes and understand key concepts After a brief review of each section's critical ideas students are taken through worked out examples try it yourself examples and chapter quizzes all structured to reinforce chapter objectives and build problem solving techniques

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