

CHAPTER 3 STUDY GUIDE FOR CONTENT MASTERY

Matter—Properties and Changes**Section 3.1 Properties of Matter**

Fill in the blank, read about physical properties and chemical properties of matter.

Use each of the terms below just once to complete the passage.

chemical density	mass properties	physical substances
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Matter is anything with **mass** and **volume**. A **substance** is a form of matter with a constant and unchanging composition. Substances have specific, unchanged **properties** that can be observed. Substances have both **physical** and **chemical** properties. **Physical** properties can be observed without changing a substance's chemical composition. Color, hardness, and **density** are examples. Other properties cannot be observed without changing the composition of a substance. These are called **chemical** properties. An example is the tendency of zinc to form rust when exposed to air.

Label each property as either physical or chemical.

physical **8.** Chemical formula H_2O chemical **9.** Turns green when exposed to airchemical **10.** Remains unchanged when in the presence of nitrogenphysical **11.** Colorlessphysical **12.** Solid at normal temperatures and pressureschemical **13.** Ability to combine with another substancephysical **14.** Melting pointphysical **15.** Liquid at normal temperatures and pressuresphysical **16.** Boiling point is 100°C physical **17.** Conducts electricityphysical **18.** Density is $\frac{1}{20}$

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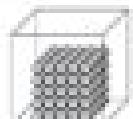
Section 3.1 continued

In your textbook, read about states of matter.

Label each drawing with one of these words: solid, liquid, gas.



gas



solid



liquid

For each statement below, write true or false.

true **19.** All matter that we encounter in everyday life exists in one of these physical forms.true **20.** A solid has definite shape and volume.false **21.** A liquid has a definite shape and takes on the volume of its container.true **22.** A gas has both the shape and the volume of its container.false **23.** The particles in a gas cannot be compressed into a smaller volume.false **24.** Liquids tend to contract when heated.false **25.** The particles in a solid are spaced far apart.false **26.** The words gas and vapor can be used interchangeably.

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**Robin M. Kowalski, Alastair
Younger, Drew Westen**

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