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web jan 29 2020 step 3 determine the number of bonds in the molecule covalent bonds are formed when one electron from each atom forms an electron pair step 2 tells how many electrons are needed and step 1 is how many electrons you have subtracting the number in step 1 from the number in step 2 gives you the number of electrons needed to complete

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web use the vsepr to determine the electronic and molecular geometries of the structure an example of drawing a lewis dot structure for nf 3 is as follows 1 determine the total number of valence electrons that must appear in the structure

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web this effect determines the atom's geometry and bond angles two regions will be 180° apart three regions will be 120° apart and four regions will be 109.5° apart geometry determination a determine the lewis dot structure of the molecule or ion b for each central atom in the structure determine the areas of electron density that lie

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